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SINCE FILE TOTAL ENTRY SESSION 0.22 0.22

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FILE CONTENT:1840 - 19 Apr 2009 VOL 150 ISS 17

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L1 STRUCTURE UPLOADED

=> d 11L1 HAS NO ANSWERS L1 STR



Structure attributes must be viewed using STN Express query preparation.

=> s 11SAMPLE SEARCH INITIATED 13:02:08 FILE 'CASREACT' SCREENING COMPLETE -778 REACTIONS TO VERIFY FROM 84 DOCUMENTS

100.0% DONE SEARCH TIME: 00.00.01

778 VERIFIED 105 HIT RXNS

19 DOCS

FULL FILE PROJECTIONS: ONLINE \*\*COMPLETE\*\*

BATCH \*\*COMPLETE\*\*

PROJECTED VERIFICATIONS: 13888 TO 17232 PROJECTED ANSWERS: 119 TO 641

L2 19 SEA SSS SAM L1 ( 105 REACTIONS)

=> s 11 full

FULL SEARCH INITIATED 13:02:12 FILE 'CASREACT'

SCREENING COMPLETE - 9335 REACTIONS TO VERIFY FROM 1553 DOCUMENTS

100.0% DONE 9335 VERIFIED 636 HIT RXNS 247 DOCS

SEARCH TIME: 00.00.04

L3 247 SEA SSS FUL L1 ( 636 REACTIONS)

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=> file caplus

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ENTRY SESSION

FULL ESTIMATED COST 123.13 123.35

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FILE COVERS 1907 - 22 Apr 2009 VOL 150 ISS 17 FILE LAST UPDATED: 21 Apr 2009 (20090421/ED)

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This file contains CAS Registry Numbers for easy and accurate substance identification.

=> s 13

L4 247 L3

=> s 14 and methanol 247957 METHANOL

L5 52 L4 AND METHANOL

=> s 15 not py > 2004

6074028 PY > 2004

L6 31 L5 NOT PY > 2004

=> d occ 16 1-31

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L6 ANSWER 1 OF 31 CAPLUS COPYRIGHT 2009 ACS on STN
FIELD COUNT
              1
AN
              1
ΙT
L6 ANSWER 2 OF 31 CAPLUS COPYRIGHT 2009 ACS on STN
FIELD COUNT
ΑN
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ΙT
L6 ANSWER 3 OF 31 CAPLUS COPYRIGHT 2009 ACS on STN
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L6 ANSWER 4 OF 31 CAPLUS COPYRIGHT 2009 ACS on STN
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L6 ANSWER 6 OF 31 CAPLUS COPYRIGHT 2009 ACS on STN
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ΙT
               1
L6 ANSWER 7 OF 31 CAPLUS COPYRIGHT 2009 ACS on STN
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AΒ
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L6 ANSWER 9 OF 31 CAPLUS COPYRIGHT 2009 ACS on STN
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L6 ANSWER 10 OF 31 CAPLUS COPYRIGHT 2009 ACS on STN
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ΙT
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L6 ANSWER 11 OF 31 CAPLUS COPYRIGHT 2009 ACS on STN
FIELD COUNT
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AB Phosphotungstic acid was used as an esterification catalyst for synthesizing di-Me fumarate (DMF), with maleic anhydride as the starting material and potassium bromate KBrO3 as the isomerizing agent. This method has the advantages of requiring small amount of catalyst with high catalysis activity, resulting in shorter reaction time and high DMF yield

```
(typically over 90%). The purification procedure of DMF is simple.
     2005:396547 CAPLUS
ΑN
     144:369644
DN
     67-56-1, Methanol, reactions
ΤТ
     RL: NUU (Other use, unclassified); RCT (Reactant); RACT (Reactant or
     reagent); USES (Uses)
        (catalytic synthesis of di-Me fumarate with phosphotungstic acid)
     ANSWER 2 OF 31 CAPLUS COPYRIGHT 2009 ACS on STN
L6
ACCESSION NUMBER:
                         2004:626179 CAPLUS
DOCUMENT NUMBER:
                         141:259007
TITLE:
                         Synthesis of chiral phosphoantigens and their activity
                         in \gamma\delta T cell stimulation
AUTHOR(S):
                         Song, Yongcheng; Zhang, Yonghui; Wang, Hong; Raker,
                         Amy; Sanders, John; Broderick, Erin; Clark, Allen;
                         Morita, Craig; Oldfield, Eric
                         Department of Chemistry, University of Illinois at
CORPORATE SOURCE:
                         Urbana-Champaign, Urbana, IL, 61801, USA
                         Bioorganic & Medicinal Chemistry Letters (2004),
SOURCE:
                         14(17), 4471-4477
                         CODEN: BMCLE8; ISSN: 0960-894X
PUBLISHER:
                         Elsevier B.V.
DOCUMENT TYPE:
                         Journal
LANGUAGE:
                         English
                         CASREACT 141:259007
OTHER SOURCE(S):
     \gamma\delta T cells expressing V\gamma2V\delta2 T cell receptors are
     activated by a broad range of phosphorus-containing small mols., termed
     phosphoantigens, and are of interest in the context of the chemotherapy of
     B cell malignancies. Here, we report the synthesis of four pairs of
     chiral phosphoantigens: the bromohydrins of isopentenyl diphosphate
     (PhosphostimTM), the epoxides of isopentenyl diphosphate (EIPP); and the
     corresponding bromohydrin and epoxide analogs of but-3-enyl diphosphate.
     The ability of each compound to stimulate human V\gamma 2V\delta 2 T cells
     was determined by TNF-\alpha release and cell proliferation. In these assays,
     the (R)-bromohydrin diphosphates, were, on average, about twice as active as
     the (S)-bromohydrin diphosphates. In contrast, the (S)-form of EIPP was
     about twice as active as (R)-EIPP. The activities of the epoxy but
     -3-enyl diphosphates were both very low. These results suggest that
     chiral phosphoantigens, as opposed to racemic mixts., may have utility in
     immunotherapy.
REFERENCE COUNT:
                         20
                               THERE ARE 20 CITED REFERENCES AVAILABLE FOR THIS
                               RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT
ΑN
     2004:626179 CAPLUS
DN
     141:259007
ΤТ
     67-56-1, Methanol, reactions
                                   124-63-0, Methanesulfonyl
                2857-97-8
                           6236-09-5
                                        6236-10-8
                                                     64028-90-6
     chloride
                                                                  69739-34-0
     76282-45-6
                76282-49-0 432544-59-7
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (synthesis of chiral phosphoantigens and their activity in
        \gamma\delta T cell stimulation)
     ANSWER 3 OF 31 CAPLUS COPYRIGHT 2009 ACS on STN
                         2004:557260 CAPLUS
ACCESSION NUMBER:
DOCUMENT NUMBER:
                         142:176440
TITLE:
                         Recovering of dibasic acid ester from waste alkali
                         liquor of caprolactam preparation
INVENTOR(S):
                         Zhou, Xianjun; Wang, Keshun; Wu, Chonghe; Liu,
                         Yaozhong; Jiang, Ping
PATENT ASSIGNEE(S):
                         Zhongxinghua Industrial Co., Ltd., Peop. Rep. China
SOURCE:
                         Faming Zhuanli Shenqing Gongkai Shuomingshu, 19 pp.
                         CODEN: CNXXEV
```

DOCUMENT TYPE:

LANGUAGE:

Patent Chinese FAMILY ACC. NUM. COUNT: 1
PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	
CN 1401624	А	20030312		
CN 1172898 IORITY APPLN. INF	0.:	20041027	CN 2001-124260	20010820
of caprolactam idizing the organic substa	method comp preparation	n to sep. ac	.40 neutralizing the wast queous phase and organ e with an oxidant sel	ic phase and
into a two-seg			reaction product of t t to recover monobasi	
segment to obt	ain crude Ca e of catalys xchange res	4-6 dibasic st, such as	ual HNO3 and nitro co acid, (3) esterifying H2SO4, H3PO4, HNO3, s illing	with C1-4 alc.
67-56-1, Metha reactions 76 permanganate RL: RCT (React (recovering	97-37-2, Nit 7722-84-1, ant); RACT	tric acid, r Hydrogen pe (Reactant or acid ester	8-2, Phosphoric acid, eactions 7722-64-7, eroxide, reactions reagent) from waste alkali lic	Potassium
ANSWER 4 OF 31 CESSION NUMBER: CUMENT NUMBER: TLE: THOR(S): RPORATE SOURCE:	2004:1 141:5 Synthe Wang, Depart	163065 CAPI 4226 esis of 2-(5 Ya-Lou; Liu tment of Med	9 ACS on STN US -methyl-2-phenyl-4-ox , Xing; Li, Jiang-Chu licinal Chemistry, Chi iversity, Nanjing, 21	ian .na
URCE:	Yingyo		2004), 21(1), 104-106 SN: 1000-0518	
L-aspartic aci Dakin-West rea overall yield.	Kexue Journa Chines CASREA phenyl-4-oxa d via 5-step ction, cycla	Chubanshe al se ACT 141:5422 azolyl)ethar reactions; izaction and L-aspartic a		enzoylation, .H4 in about 31.2% 0
was benzoylate (II). Dakin-W which was cycl 2-(5-methyl-2- I by treatment compds. were c 2004:163065 C 141:54226 2-(5-Methyl-2- L-aspartic aci Dakin-West rea	d to give 82 est reaction ized in toluphenyl-4-oxawith ether haracterized APLUS  phenyl-4-oxad via 5-sterction, cycli	2.0% N-benzon of II gave uene by POClazolyl) aceta solution of by 1H NMR, azolyl) ethar oreactions;	yl-L-aspartic acid β- 274.4% Me 3-benzamido 3 to give 81.8% Me ate, the latter can ea LiAlH4 in yield of 8 IR and elemental ana acid (I) was synthesized esterification, N-bell reduction using LiAll acid with methanol in	Me ester 0-4-oxovalerate, esily convert to 16%. All the 11.  ed from enzoylation, H4 in about 31.2% 0

°C gave 72.6% L-aspartic acid  $\beta\text{-Me}$  ester hydrochloride, which was benzoylated to give 82.0% N-benzoyl-L-aspartic acid  $\beta\text{-Me}$  ester

(II). Dakin-West reaction of II gave 74.4% Me 3-benzamido-4-oxovalerate, which was cyclized in toluene by POC13 to give 81.8% Me 2-(5-methyl-2-phenyl-4-oxazolyl) acetate, the latter can easily convert to I by treatment with ether solution of LiAlH4 in yield of 86%. All the compds. were characterized by 1H NMR, IR and elemental anal.

L6 ANSWER 5 OF 31 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 2003:954590 CAPLUS

DOCUMENT NUMBER: 141:191003

TITLE: Reactions of 2H-3,1-benzoxazine -2,4(1H)-dione

AUTHOR(S): Milea, M.; Csunderlik, C.; Fota, Daniela

CORPORATE SOURCE: Department of Organic Chemistry, The University "POLITEHNICA" Timisoara, Timisoara, RO-1900, Rom.

SOURCE: Buletinul Stiintific al Universitatii "Politehnica"

din Timisoara Romania, Seria Chimie si Mediului

(2000), 45(1), 163-167

CODEN: BSIMFG; ISSN: 1224-6018

PUBLISHER: Universitatii "Politehnica" din Timisoara

DOCUMENT TYPE: Journal LANGUAGE: English

OTHER SOURCE(S): CASREACT 141:191003

AB Reactions of isatoic anhydride (2H-3,1-benzoxazine -2,4(1H)-dione) with alcs. and phenols, both in the presence or absence of basic catalysts, have been studied, conditions have been established and the products characterized by physico-chemical methods. Also the reactions of isatoic anhydride with Me esters of  $\alpha$ -amino acids were studied among the reactions of isatoic anhydride with nitrogen nucleophiles.

REFERENCE COUNT: 5 THERE ARE 5 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

AN 2003:954590 CAPLUS

DN 141:191003

ST isatoic anhydride benzoxazinedione reaction alc phenol; amino acid ester nucleophile prepn esterification methanol benzoxazinedione reaction

IT Amino acids, preparation

RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)

(esters; prepn of Me esters of amino acids by esterification of amino acids with methanol)

IT Esterification

(prepn of Me esters of amino acids by esterification of amino acids with methanol)  $\$ 

IT 67-56-1, Methanol, reactions

RL: RCT (Reactant); RACT (Reactant or reagent)

(prepn of Me esters of amino acids by esterification of amino acids with methanol)

IT 5680-79-5P 6384-18-5P

RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)

(prepn of Me esters of amino acids by esterification of amino acids with methanol)  $\$ 

IT 2491-18-1P 7517-19-3P

RL: SPN (Synthetic preparation); PREP (Preparation)

(prepn of Me esters of amino acids by esterification of amino acids with methanol)

L6 ANSWER 6 OF 31 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 2003:363327 CAPLUS

DOCUMENT NUMBER: 139:230335

TITLE:

Base-induced alcoholysis of N-arylmaleimides: facile
in situ oxa-michael addition to alkyl maleanilates:
Two-step one-pot rapid access to alkoxy-succinic acids

AUTHOR(S): Mhaske, Santosh B.; Argade, Narshinha P.

CORPORATE SOURCE: Division of Organic Chemistry (Synthesis), National

Chemical Laboratory, Pune, 411 008, India

Synthesis (2003), (6), 863-870SOURCE:

CODEN: SYNTBF; ISSN: 0039-7881

Georg Thieme Verlag PUBLISHER:

DOCUMENT TYPE: Journal English LANGUAGE:

OTHER SOURCE(S): CASREACT 139:230335

A simple, efficient and general two-step, one-pot approach to alkoxysuccinic acids is described. The potassium carbonate-catalyzed reactions of alcs. with 1-(4-methylphenyl)-1H-pyrrole-2,5-dione followed by an acid-induced hydrolysis of intermediate products furnished alkoxysuccinic acids in 90-98% yields. All the intermediates from the reaction of 1-(4-Methylphenyl)-1H-Pyrrole-2,5-dione were characterized, proving that the in situ formed alkyl maleanilates are the actual Michael acceptors.

REFERENCE COUNT: 61 THERE ARE 61 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

2003:363327 CAPLUS ΑN

139:230335 DN

64-17-5, Ethanol, reactions 67-56-1, Methanol, reactions ΤT

67-63-0, 2-Propanol, reactions 71-41-0, 1-Pentanol, reactions

100-51-6, Benzenemethanol, reactions 107-21-1, 1,2-Ethanediol, reactions 108-31-6, Maleic anhydride, reactions 111-87-5, 1-Octanol, reactions

112-30-1, 1-Decanol 112-53-8, 1-Dodecanol 3886-69-9,

24870-11-9,  $(\alpha R) - \alpha$ -Methylbenzenemethanamine

(2Z)-4-[(4-Methylphenyl)amino]-4-oxo-2-butenoic acid

RL: RCT (Reactant); RACT (Reactant or reagent)

(two-step one-pot preparation of (alkoxy) succinic acids via base-induced alcoholysis of N-arylmaleimides and facile in situ oxa-Michael addition to alkyl maleanilates)

ANSWER 7 OF 31 CAPLUS COPYRIGHT 2009 ACS on STN

2003:44804 CAPLUS ACCESSION NUMBER:

DOCUMENT NUMBER: 138:337689

TITLE: Synthesis of dimethyl fumarate by heterogeneous

supported heteropoly acid

Xu, Wenyuan; Peng, Daofeng; Xiong, Guoxuan; Zhu, AUTHOR(S):

Xiaping

Department of Applied Chemistry, East China Institute CORPORATE SOURCE:

of Technology, Fuzhou, 344000, Peop. Rep. China

SOURCE: Huaxue Shiji (2002), 24(6), 367-368

> CODEN: HUSHDR; ISSN: 0258-3283 Huagongbu Huaxue Shiji Xinsizhan

DOCUMENT TYPE: Journal LANGUAGE: Chinese

CASREACT 138:337689 OTHER SOURCE(S):

Synthesis of di-Me fumarate by esterification reaction of fumaric acid with methanol catalyzed by heterogeneous supported heteropoly acid PW12/C was studied in this paper. A careful study of the effects on the esterification reaction was done. Under these conditions, the yield of ester was about 91.6%.

2003:44804 CAPLUS ΑN

DN138:337689

PUBLISHER:

Synthesis of di-Me fumarate by esterification reaction of fumaric acid with methanol catalyzed by heterogeneous supported heteropoly acid PW12/C was studied in this paper. A careful study of the effects on the esterification reaction was done. Under these conditions, the yield of ester was about 91.6%.

110-17-8, Fumaric acid, reactions 67-56-1, Methanol, reactions RL: RCT (Reactant); RACT (Reactant or reagent)

(synthesis of di-Me fumarate by heterogeneous supported heteropoly acid)

```
ANSWER 8 OF 31 CAPLUS COPYRIGHT 2009 ACS on STN
1.6
                         2002:952405 CAPLUS
ACCESSION NUMBER:
                         139:6591
DOCUMENT NUMBER:
                         Catalytic synthesis of dimethyl fumarate using
TITLE:
                         solid-supported superacid catalyst
                         Zhao, Lifang; He, Zhusheng; Ma, Yuying
AUTHOR(S):
                         Dept. Chem .+ Chem. Eng., Baoji Coll. Arts + Sci.,
CORPORATE SOURCE:
                         Baoji, 721007, Peop. Rep. China
SOURCE:
                         Baoji Wenli Xueyuan Xuebao, Ziran Kexueban (2002),
                         22(2), 138-140
                         CODEN: BWZKFL
PUBLISHER:
                         Baoji Wenli Xueyuan Xuebao Bianjibu
DOCUMENT TYPE:
                         Journal
LANGUAGE:
                         Chinese
                         CASREACT 139:6591
OTHER SOURCE(S):
     The preparation of supported catalyst, TiO2/La3+/SO42- supported on mol.
     sieves, and its catalytic activity to esterification of fumarate were
     studied. The catalyst had fine catalytic activity. The optimum
     conditions of the esterification were decided by orthogonal expts. as
     follows: activation temperature of the catalyst was 500°, the amount of
     catalyst was 15% (based on the mass of fumaric acid), the mole ratio of
     alc. to acid was 6:1 and the reaction time was 5 h. Under the optimum
     reaction conditions, the yield of di-Me fumarate was up to 92.3%.
ΑN
     2002:952405 CAPLUS
     139:6591
DN
     67-56-1, Methanol, reactions
                                    110-17-8, Fumaric acid, reactions
ΤT
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (synthesis of di-Me fumarate using solid-supported superacid catalyst)
     ANSWER 9 OF 31 CAPLUS COPYRIGHT 2009 ACS on STN
ACCESSION NUMBER:
                         2002:903288 CAPLUS
DOCUMENT NUMBER:
                         138:271016
TITLE:
                         A simple, convenient and expeditious route to methyl
                         esters of carboxylic acids by thionyl chloride-
                         methanol
AUTHOR(S):
                         Chatterjee, Tapasi; Chattopadhyay, Subhagata
CORPORATE SOURCE:
                         Department of Chemistry, Jadavpur University, Kolkata,
                         700 032, India
SOURCE:
                         Oriental Journal of Chemistry (2002), 18(2), 187-190
                         CODEN: OJCHEG; ISSN: 0970-020X
PUBLISHER:
                         Oriental Scientific Publishing Co.
DOCUMENT TYPE:
                         Journal
LANGUAGE:
                        English
OTHER SOURCE(S):
                         CASREACT 138:271016
     A simple, convenient and expeditious preparation of 40-90\% Me esters of
     carboxylic acids by thionyl chloride and MeOH was described. Among the 29
     esters prepared were 90% 2-IC6H4CO2Me, 87% 4-MeOC6H4CO2Me and 86%
     Bz(CH2)2CO2Me.
                               THERE ARE 15 CITED REFERENCES AVAILABLE FOR THIS
REFERENCE COUNT:
                         15
                               RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT
     2002:903288 CAPLUS
AN
DN
     138:271016
ΤI
     A simple, convenient and expeditious route to methyl esters of carboxylic
     acids by thionyl chloride-methanol
     thionyl chloride methanol esterification carboxylic acid
ΙT
     Esterification
        (preparation of Me esters of carboxylic acids by thionyl chloride-
        methanol)
ΤT
     Esters, preparation
     RL: SPN (Synthetic preparation); PREP (Preparation)
```

(preparation of Me esters of carboxylic acids by thionyl chloride-

methanol)

```
2-Hydroxybenzoic acid, reactions 79-11-8, Chloroacetic acid, reactions
     87-69-4, L-Tartaric acid, reactions 88-67-5, 2-Iodobenzoic acid
     88-99-3, Phthalic acid, reactions 99-06-9, 3-Hydroxybenzoic acid,
                99-34-3, 3,5-Dinitrobenzoic acid 99-96-7, 4-Hydroxybenzoic
     reactions
     acid, reactions 100-09-4, 4-Methoxybenzoic acid 103-82-2, Phenylacetic
     acid, reactions 110-15-6, Succinic acid, reactions 110-16-7, Maleic
     acid, reactions 110-17-8, Fumaric acid, reactions 110-44-1,
     2,4-Hexadienoic acid 117-34-0, Diphenylacetic acid 118-91-2,
     2-Chlorobenzoic acid 121-92-6, 3-Nitrobenzoic acid 140-10-3,
     trans-Cinnamic acid, reactions 141-82-2, Malonic acid, reactions
     144-62-7, Oxalic acid, reactions 527-72-0, 2-Thiophenecarboxylic acid
     552-16-9, 2-Nitrobenzoic acid 2051-95-8, 3-Phenylpropionic acid
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (preparation of Me esters of carboxylic acids by thionyl chloride-
        methanol)
     64-17-5, Ethanol, reactions 67-56-1, Methanol, reactions
ΤT
     67-63-0, Isopropanol, reactions 7719-09-7, Thionyl chloride
     RL: RGT (Reagent); RACT (Reactant or reagent)
        (preparation of Me esters of carboxylic acids by thionyl chloride-
       methanol)
ΤT
     93-58-3P, Methyl benzoate 93-89-0P, Ethyl benzoate 96-34-4P, Methyl
     chloroacetate 99-76-3P, Methyl 4-hydroxybenzoate 99-77-4P, Ethyl 2-nitrobenzoate 101-41-7P, Methyl phenylacetate 106-65-0P, Dimethyl
     succinate 108-59-8P, Dimethyl malonate 119-36-8P, Methyl
     2-hydroxybenzoate 121-98-2P, Methyl 4-methoxybenzoate
                                                             131-11-3P,
     Dimethyl phthalate 553-90-2P, Dimethyl oxalate 606-27-9P, Methyl
     2-nitrobenzoate 608-68-4P 610-96-8P, Methyl 2-chlorobenzoate
     610-97-9P, Methyl 2-iodobenzoate 618-95-1P, Methyl 3-nitrobenzoate
     619-50-1P, Methyl 4-nitrobenzoate 624-49-7P, Dimethyl fumarate
     689-89-4P 939-48-0P, Isopropyl benzoate 1754-62-7P
                                                             2702-58-1P,
    Methyl 3,5-dinitrobenzoate 3469-00-9P, Methyl diphenylacetate
     5380-42-7P, Methyl 2-thiophenecarboxylate 13756-40-6P, Isopropyl
                     19438-10-9P, Methyl 3-hydroxybenzoate 25333-24-8P,
     2-nitrobenzoate
     Methyl 3-benzoylpropionate
     RL: SPN (Synthetic preparation); PREP (Preparation)
        (preparation of Me esters of carboxylic acids by thionyl chloride-
       methanol)
    ANSWER 10 OF 31 CAPLUS COPYRIGHT 2009 ACS on STN
ACCESSION NUMBER:
                        2002:81290 CAPLUS
DOCUMENT NUMBER:
                        137:352688
TITLE:
                        Catalytic reaction-distillation synthesis of dimethyl
                        fumarate by fixed-carried heteropoly acid
AUTHOR(S):
                        Ding, Bin; Guo, Xiangming
                        Jilin Institute of Chemical Technology, Jilin,
CORPORATE SOURCE:
                        1320022, Peop. Rep. China
SOURCE:
                        Dongbei Shida Xuebao, Ziran Kexueban (2001), 33(4),
                        61-65
                        CODEN: DSZKEE; ISSN: 1000-1832
                        Dongbei Shifan Daxue Xueshu Qikanshe
PUBLISHER:
DOCUMENT TYPE:
                        Journal
LANGUAGE:
                        Chinese
                        CASREACT 137:352688
OTHER SOURCE(S):
     A new synthesis technol. of di-Me fumarate was presented. Fumarate,
     methanol, and self-made fixed-carried heteropoly acid as catalyst
     were used. The reaction-distillation conditions were ratio of alc. and acid
     about 7:1; esterification temperature about 67-78^{\circ}; and reaction time
     \leq6 h. The yield of product was up to 92%.
     2002:81290 CAPLUS
ΑN
    137:352688
DN
    A new synthesis technol. of di-Me fumarate was presented. Fumarate,
AB
    methanol, and self-made fixed-carried heteropoly acid as catalyst
```

62-23-7, 4-Nitrobenzoic acid 65-85-0, Benzoic acid, reactions

ΙT

69-72-7,

were used. The reaction-distillation conditions were ratio of alc. and acid about 7:1; esterification temperature about  $67-78^{\circ}$ ; and reaction time

≤6 h. The yield of product was up to 92%.

67-56-1, Methanol, reactions 110-17-8, Fumaric acid, reactions

RL: RCT (Reactant); RACT (Reactant or reagent)

(catalytic reaction-distillation synthesis of di-Me fumarate by fixed-carried

heteropoly acid)

ANSWER 11 OF 31 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 2001:537078 CAPLUS

DOCUMENT NUMBER: 135:256190

TITLE: Lipase-catalyzed synthesis of RGD diamide in aqueous

water-miscible organic solvents

Zhang, L.-Q.; Zhang, Y.-D.; Xu, L.; Li, X.-L.; Yang, AUTHOR(S):

X.-c.; Xu, G.-L.; Wu, X.-X.; Gao, H.-Y.; Du, W.-B.;

Zhang, X.-T.; Zhang, X.-Z.

Key Laboratory of Molecular Enzymology and CORPORATE SOURCE:

Engineering, Jilin University, Changchun, 130023,

Peop. Rep. China

SOURCE: Enzyme and Microbial Technology (2001), 29(2-3),

129-135

CODEN: EMTED2; ISSN: 0141-0229 Elsevier Science Ireland Ltd.

DOCUMENT TYPE: Journal LANGUAGE: English

CASREACT 135:256190 OTHER SOURCE(S):

1RGD tripeptide as the cellular adhesion factor was synthesized by a combination of chemical and enzymic methods in this study. First of all, Gly-Asp diamide was synthesized by a novel chemical method in three steps including preparation of L-aspartic acid di-Me ester, chloroacetylation of L-aspartic acid di-Me ester and ammonolysis of chloroacetyl L-aspartic acid di-Me ester. Secondly, Porcine Pancreas Lipase (PPL) was used to catalyze the synthesis of Benzyl-Arg-Gly-Asp diamide in water-organic cosolvents systems. The reaction condition was optimized by examining the main factors affecting the yield of the tripeptide. The optimal reaction condition was set up as pH7.7,  $15^{\circ}$ C in 50% DMF for 8 h with the maximum yield of 76.4%. It was also found that 50% DMSO was another alternative with the tripeptide yield of 71.7%.

REFERENCE COUNT: 27 THERE ARE 27 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

ΑN 2001:537078 CAPLUS

DN 135:256190

PUBLISHER:

ΙT 56-84-8, L-Aspartic acid, reactions 67-56-1, Methanol,

reactions 79-04-9, Chloroacetyl chloride 1336-21-6, Ammonium hydroxide 7664-41-7, Ammonia, reactions

RL: RCT (Reactant); RACT (Reactant or reagent)

(lipase-catalyzed synthesis of RGD diamide in aqueous water-miscible organic solvents)

ANSWER 12 OF 31 CAPLUS COPYRIGHT 2009 ACS on STN

2001:535025 CAPLUS ACCESSION NUMBER:

136:294520 DOCUMENT NUMBER:

TITLE: Synthesis of dimethyl fumarate catalyzed by SO42-/TiO2/La3+ rare earth solid superacid

AUTHOR(S): Zhou, Jianwei

CORPORATE SOURCE: Department of Chemical Engineering, Pingyuan University, Xinxiang, 453003, Peop. Rep. China

Henan Huagong (2001), (5), 12-14 CODEN: HEHUF3; ISSN: 1003-3467 SOURCE:

Henansheng Shiyou Huaxue Gongye Keji Qingbao PUBLISHER:

Zhongxinzhan

DOCUMENT TYPE: Journal LANGUAGE: Chinese

OTHER SOURCE(S): CASREACT 136:294520

AB Di-Me fumarate was synthesized from fumaric acid and methanol with SO42-/TiO2/La3+ rare earth solid superacid as catalyst in dichloromethane solvent. Optimum synthetic conditions were determined: molar ratio of fumaric acid to methanol 6:1, dosage of catalyst 1.0 g/0.1 mol fumaric acid, time 4 h and solvent 25 mL. Yield of product reached above 94%.

AN 2001:535025 CAPLUS

DN 136:294520

AB Di-Me fumarate was synthesized from fumaric acid and methanol with SO42-/TiO2/La3+ rare earth solid superacid as catalyst in dichloromethane solvent. Optimum synthetic conditions were determined: molar ratio of fumaric acid to methanol 6:1, dosage of catalyst 1.0 g/0.1 mol fumaric acid, time 4 h and solvent 25 mL. Yield of product reached above 94%.

ST dimethyl fumarate synthesis fumaric acid methanol solid superacid catalysis

L6 ANSWER 13 OF 31 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 2001:461290 CAPLUS

DOCUMENT NUMBER: 136:279092

TITLE: Synthesis of dimethyl fumarate from maleic acid

AUTHOR(S): Cao, Kelin

CORPORATE SOURCE: Shanxi Taiming Chemical Engineering Co., Ltd., Taigu,

030800, Peop. Rep. China

SOURCE: Huagong Jinzhan (2001), 20(4), 33-34, 39

CODEN: HUJIEK; ISSN: 1000-6613

PUBLISHER: Huaxue Gongye Chubanshe

DOCUMENT TYPE: Journal LANGUAGE: Chinese

OTHER SOURCE(S): CASREACT 136:279092

AB Title compound was prepared from maleic acid, isomerized fumaric acid in the presence of ammonium persulfateto as catalyst, further esterification with methanol in the presence of phosphotungstic acid as catalyst, giving product with yield over 94%. The effects of catalysts and catalyst amount on the reactions were studied.

AN 2001:461290 CAPLUS

DN 136:279092

AB Title compound was prepared from maleic acid, isomerized fumaric acid in the presence of ammonium persulfateto as catalyst, further esterification with methanol in the presence of phosphotungstic acid as catalyst, giving product with yield over 94%. The effects of catalysts and catalyst amount on the reactions were studied.

L6 ANSWER 14 OF 31 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 2001:443163 CAPLUS

DOCUMENT NUMBER: 136:263409

TITLE: Synthesis of new chiral lipophilic macrocyclic oxo

polyamines

AUTHOR(S): Yan, Qianshun; You, Jinsong; Xiang, Qingxiang; Yu,

Xiaoqi; Xie, Rugang

CORPORATE SOURCE: Department of Chemistry, Sichuan University, Chengdu,

610064, Peop. Rep. China

SOURCE: Huaxue Yanjiu Yu Yingyong (2001), 13(2), 189-191

CODEN: HYYIFM; ISSN: 1004-1656

PUBLISHER: Huaxue Yanjiu Yu Yingyong Bianjibu

DOCUMENT TYPE: Journal LANGUAGE: Chinese

OTHER SOURCE(S): CASREACT 136:263409

AB New chiral lipophilic macrocyclic oxo polyamines were synthesized from L-glutamic acid or L-aspartic acid by esterifying with methanol, acylating with dodecanoyl chloride in chloroform in the presence of K2CO3 to obtain di-Me N-dodecanoyl-L-glutamate or L-aspartate, and transacylating with diethylenetriamine or triethylenetetraamine in methanol under refluxing. Their structures were characterized by MS, 1H NMR, and elemental anal.

AN 2001:443163 CAPLUS

DN 136:263409

AB New chiral lipophilic macrocyclic oxo polyamines were synthesized from L-glutamic acid or L-aspartic acid by esterifying with methanol, acylating with dodecanoyl chloride in chloroform in the presence of K2CO3 to obtain di-Me N-dodecanoyl-L-glutamate or L-aspartate, and transacylating with diethylenetriamine or triethylenetetraamine in methanol under refluxing. Their structures were characterized by MS, 1H NMR, and elemental anal.

L6 ANSWER 15 OF 31 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 2000:832322 CAPLUS

DOCUMENT NUMBER: 134:310893

TITLE: Synthesis of dimethyl fumarate catalyzed by composite

solid superacid SO42-/TiO2-Al2O3

AUTHOR(S): Cheng, Yonghao

CORPORATE SOURCE: Department of Chemistry, Hebei Normal University,

Shijiazhuang, 050016, Peop. Rep. China

SOURCE: Riyong Huaxue Gongye (2000), 30(5), 12-13

CODEN: RHGOE8; ISSN: 1001-1803

PUBLISHER: Qinggongyebu Kexue Jishu Qingbao Yanjiuso

DOCUMENT TYPE: Journal LANGUAGE: Chinese

OTHER SOURCE(S): CASREACT 134:310893

AB Di-Me fumarate was synthesized from fumaric acid and methanol with composite solid superacid SO42-/TiO2- Al2O3 as catalyst. Optimum synthetic conditions were determined: methanol:fumaric acid 6:1, time 4 h, and dosage of catalyst 3 g. Yield of product reached 91.4%.

AN 2000:832322 CAPLUS

DN 134:310893

AB Di-Me fumarate was synthesized from fumaric acid and methanol with composite solid superacid SO42-/TiO2- Al2O3 as catalyst. Optimum synthetic conditions were determined: methanol:fumaric acid 6:1, time 4 h, and dosage of catalyst 3 g. Yield of product reached 91.4%.

IT 67-56-1, Methanol, reactions 110-17-8, Fumaric acid, reactions RL: RCT (Reactant); RACT (Reactant or reagent) (synthesis of di-Me fumarate catalyzed by composite solid superacid

SO42-/TiO2-A12O3)

L6 ANSWER 16 OF 31 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 2000:586307 CAPLUS

DOCUMENT NUMBER: 133:296252

TITLE: A simple method for the preparation of monomethyl

esters of dicarboxylic acids by selective

esterification of the nonconjugated carboxyl group in the presence of an aromatic or conjugated carboxyl  $\,$ 

group

AUTHOR(S): Ram, Ram N.; Meher, Nabin Kumar

CORPORATE SOURCE: Department of Chemistry, Indian Institute of

Technology, New Delhi, 110016, India

SOURCE: Journal of Chemical Research, Synopses (2000), (6),

282-283

CODEN: JRPSDC; ISSN: 0308-2342

PUBLISHER: Science Reviews Ltd.

DOCUMENT TYPE: Journal LANGUAGE: English

OTHER SOURCE(S): CASREACT 133:296252

AB Various dicarboxylic acids were converted selectively into monomethyl esters in which the nonconjugated carboxyl group is selectively esterified in the present of an aromatic or conjugated carboxyl group at room temperature

(.apprx.25-27°) in MeOH using a catalytic amount of SOC12.

REFERENCE COUNT: 16 THERE ARE 16 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

AN 2000:586307 CAPLUS

DN 133:296252

ST dicarboxylate methanol selective esterification; ester dicarboxylic selective prepn

L6 ANSWER 17 OF 31 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1998:732350 CAPLUS

DOCUMENT NUMBER: 130:66049

TITLE: A selective method for the preparation of aliphatic

methyl esters in the presence of aromatic carboxylic

acids

AUTHOR(S): Rodriguez, A.; Nomen, M.; Spur, B. W.

CORPORATE SOURCE: Department of Cell Biology, Univ. of Medicine and

Dentistry of New Jersey, Stratford, NJ, 08084, USA

SOURCE: Tetrahedron Letters (1998), 39(47), 8563-8566

CODEN: TELEAY; ISSN: 0040-4039

PUBLISHER: Elsevier Science Ltd.

DOCUMENT TYPE: Journal LANGUAGE: English

OTHER SOURCE(S): CASREACT 130:66049

AB 2,2-Dimethoxypropane, methanol and a catalytic amount of HCl

selectively esterify aliphatic carboxylic acids in the presence of aromatic

carboxylic acids at room temperature and in high yields.

REFERENCE COUNT: 8 THERE ARE 8 CITED REFERENCES AVAILABLE FOR THIS

RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

AN 1998:732350 CAPLUS

DN 130:66049

AB 2,2-Dimethoxypropane, methanol and a catalytic amount of HCl selectively esterify aliphatic carboxylic acids in the presence of aromatic carboxylic acids at room temperature and in high yields.

L6 ANSWER 18 OF 31 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1994:680006 CAPLUS

DOCUMENT NUMBER: 121:280006

ORIGINAL REFERENCE NO.: 121:51115a,51118a

TITLE: A convenient and mild procedure for the preparation of

hydroxy esters from lactones and hydroxy acids

AUTHOR(S): Anand, R. C.; Selvapalam, N.

CORPORATE SOURCE: Dep. Chem., Indian Inst. Technol., New Delhi, 110016,

India

SOURCE: Synthetic Communications (1994), 24(19), 2743-7

CODEN: SYNCAV; ISSN: 0039-7911

DOCUMENT TYPE: Journal LANGUAGE: English

OTHER SOURCE(S): CASREACT 121:280006

AB An alc. solution of lactones and hydroxy acids stored on Amberlyst-15 results in the formation of corresponding hydroxy esters in high yields.

AN 1994:680006 CAPLUS

DN 121:280006

OREF 121:51115a,51118a

IT 64-17-5, Ethanol, reactions 67-56-1, Methanol, reactions

79-14-1, reactions 87-41-2, 1(3H)-Isobenzofuranone 87-69-4, reactions

96-48-0 119-84-6 502-44-3, 2-Oxepanone 542-28-9 553-86-6,

2(3H)-Benzofuranone 828-01-3 4026-18-0

RL: RCT (Reactant); RACT (Reactant or reagent)

(mild procedure for the preparation of hydroxy esters from lactones and hydroxy acids)

L6 ANSWER 19 OF 31 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1994:269626 CAPLUS

DOCUMENT NUMBER: 120:269626

ORIGINAL REFERENCE NO.: 120:47747a,47750a

TITLE: Catalytic synthesis of dimethyl fumarate with ferric

chloride

AUTHOR(S): Yu, Shanxin; Lei, Huanwen

CORPORATE SOURCE: Dep. Chem., Hunan Norm. UNiv., Changsha, 410081, Peop.

Rep. China

SOURCE: Huaxue Shiji (1993), 15(6), 374, 376

CODEN: HUSHDR; ISSN: 0258-3283

DOCUMENT TYPE: Journal LANGUAGE: Chinese

OTHER SOURCE(S): CASREACT 120:269626

AB Ferric chloride (FeCl3·6H2O) can be used as a catalyst for the esterification reaction of fumaric acid instead of sulfuric acid. The conditions in synthesis of di-Me fumarate catalyzed with FeCl3·6H2O are described. The advantages of this method are: simple procedure, mild reaction conditions, non-corrosive, less pollution and purer product.

AN 1994:269626 CAPLUS

DN 120:269626

OREF 120:47747a,47750a

IT Esterification catalysts

(ferric chloride, for fumaric acid with methanol)

L6 ANSWER 20 OF 31 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1993:626419 CAPLUS

DOCUMENT NUMBER: 119:226419

ORIGINAL REFERENCE NO.: 119:40439a, 40442a

TITLE: Method for preparing amino acid esters involving

(continuous) addition and distillation of alcohols

INVENTOR(S): Takemoto, Tadashi; Takeda, Hideo

PATENT ASSIGNEE(S): Ajinomoto Co., Inc., Japan SOURCE: Eur. Pat. Appl., 4 pp.

CODEN: EPXXDW

DOCUMENT TYPE: Patent LANGUAGE: English

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATE:	NO.		KINI	D DATE	A	APF	PLICATION NO.		DATE
					_				
EP 5	44205		A2	19930602	E	ΞP	1992-119898		19921123
EP 5	44205		A3	19930728					
EP 5	44205		B1	19950906					
	R: BE,	CH, D	E, FR,	GB, IT, LI,	NL				
JP 0	5148199		A	19930615	J	JΡ	1991-314585		19911128
CA 2	084012		A1	19930529	C	CA	1992-2084012		19921127
US 5	424476		A	19950613	U	JS	1994-194635		19940210
PRIORITY .	APPLN.	INFO.:			J	JΡ	1991-314585	A	19911128
					U	JS	1992-982123	В1	19921125

OTHER SOURCE(S): CASREACT 119:226419

AB Amino acid esters were prepared by heating a mixture of an amino acid, an alc., and H2SO4 with simultaneous addition of liquid or gaseous alc. to the reaction mixture and distillation of the same. Thus, H-Phe-OH in MeOH containing cat.

H2SO4 was kept at 85° for 4 h with simultaneous addition and distillation of

MeOH to give 98.6% H-Phe-OMe. Simple reflux of the above reactants gave a yield of only 83-2% product.

AN 1993:626419 CAPLUS

DN 119:226419

OREF 119:40439a,40442a

IT 56-41-7P, L-Alanine, reactions 56-84-8P, H-Asp-OH, reactions 63-91-2P, L-Phenylalanine, reactions 72-18-4P, L-Valine, reactions

RL: RCT (Reactant); PREP (Preparation); RACT (Reactant or reagent) (esterification of, with methanol, continuous distillation and addition of alc. in)

L6 ANSWER 21 OF 31 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1993:580649 CAPLUS

DOCUMENT NUMBER: 119:180649

ORIGINAL REFERENCE NO.: 119:32287a,32290a

TITLE: Preparation of pyrrolidine derivatives from amino

acids.

INVENTOR(S): Hirabayashi, Satoshi; Ike, Kazuo; Zanka, Atuhiko;

Kawakami, Takeshi; Ichihara, Masaharu

PATENT ASSIGNEE(S): Fujisawa Pharmaceutical Co., Ltd., Japan

SOURCE: PCT Int. Appl., 20 pp.

CODEN: PIXXD2

DOCUMENT TYPE: Patent LANGUAGE: Japanese

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
WO 9220652	A1	19921126	WO 1992-JP648	19920520

W: JP, US

RW: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LU, MC, NL, SE

JP 04346971 A 19921202 JP 1991-219431 19910521 PRIORITY APPLN. INFO.: JP 1991-219431 A 19910521

OTHER SOURCE(S): CASREACT 119:180649; MARPAT 119:180649

GΙ

AB Reaction of R2-NH-CH(CO2-R5)-CH2-CO2-R4 (R2, R4, R5 = protecting group) with R1-X [R1 = alkyl; X = leaving group) gives R2-NH-CH(CO2-R5)-CH(R1)-CO2-R4, whose reduction gives R2-NH-CH(CH2OH)-CH(R1)-CH2OH, whose condensation with H2N-R3 (R3 = protecting group) gives pyrrolidine derivs. I, which are optionally deprotected. Di-Me (S)-aspartate hydrochloride (preparation given) in CH2Cl2 was mixed with a saturated aqueous solution of NaHCO3, tert-di-Bu dicarbonate was

added, and the resulting mixture was allowed to stand overnight to give di-Me (S)-N-tert-butoxycarbonylaspartate, which was treated with MeI in THF containing hexamethyldisilazane and BuLi at -40 to  $-30^{\circ}$  for 2 h to give, after adjusting to pH 7-8 with 1N HCl, a 1:1 mixture of threo- and erythro-(2S)-N-(tert-butoxycarbonyl)-3-methylaspartic acid di-Me ester, which was reduced with NaBH4 in THF-MeOH at 35-50° for 4 h to give a diastereomeric mixture of (2S)-N-(tert-butoxycarbonylamino)-3-methyl-1,4-butanediol, whose dimesylate ester was heated with benzylamine at

 $30\text{--}50^{\circ}$  for 3 days to give 74% a diastereomeric mixture of

1-benzyl-3-tert-butoxycarbonylamino-4-methylpyrrolidine.

(3S,4S)-1-benzyl-3-tert-butoxycarbonylamino-4-methylpyrrolidine in

MeOH-H2O containing HCO2NH4 was hydrogenolyzed over Pd/C to give (3S,4S)-3-tert-butoxycarbonylamino-4-methylpyrrolidine, which in

MeOH-EtOAc was treated with HCl in EtOAc at room temperature for 2 h and then

at

35° for 2 h to give (3S, 4S)-3-amino-4-methylpyrrolidine.

REFERENCE COUNT: 2 THERE ARE 2 CITED REFERENCES AVAILABLE FOR THIS

RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

AN 1993:580649 CAPLUS

DN 119:180649

OREF 119:32287a,32290a

IT 56-84-8, (S)-Aspartic acid, reactions

RL: RCT (Reactant); RACT (Reactant or reagent)

(esterification of, with methanol)

L6 ANSWER 22 OF 31 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1992:612978 CAPLUS

DOCUMENT NUMBER: 117:212978

ORIGINAL REFERENCE NO.: 117:36823a,36826a

TITLE: Process for preparing diketopiperazine derivatives

INVENTOR(S): Toshihisa, Kato; Tadashi, Takemoto

PATENT ASSIGNEE(S): Ajinomoto Co., Inc., Japan

SOURCE: Eur. Pat. Appl., 5 pp.

CODEN: EPXXDW

DOCUMENT TYPE: Patent LANGUAGE: English

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.		DATE
				_	
EP 493812	A1	19920708	EP 1991-122307		19911227
R: BE, DE, FR,	GB, NL				
JP 04234374	A	19920824	JP 1990-418592		19901227
PRIORITY APPLN. INFO.:			JP 1990-418592	Α	19901227
OTHER SOURCE(S):	CASREA	CT 117:21297	8; MARPAT 117:212978		
GI					

Diketopiperazines I (R = C1-6-alkyl) were prepared by treating (chloroacetyl)aspartic acid with NH3, esterifying the resulting H-Gly-Asp-OH with an alkanol, and cyclizing the resulting diester neutral or weakly basic water or in an alkanol-water solvent mixture Thus, L-aspartic acid was acylated with ClCH2COCl in aqueous NaOH to give 62% ClCH2CO-L-Asp-OH, which was treated NH3 in water to H-Gly-Asp-OH. The latter was esterified with MeOH containing HCl to give 87% H-Gly-Asp(OMe)-OMe, which was cyclized in water at pH 7.0 at 60° for 3 h to give 72% I (R = Me).

AN 1992:612978 CAPLUS

DN 117:212978

OREF 117:36823a,36826a

IT 4685-12-5P

RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)

(preparation and esterification of, with methanol)

L6 ANSWER 23 OF 31 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1992:147458 CAPLUS

DOCUMENT NUMBER: 116:147458

ORIGINAL REFERENCE NO.: 116:24825a,24828a

TITLE: Analysis of keto acids as their methyl esters of 2,4-dinitrophenylhydrazone derivatives by gas

chromatography and gas chromatography-mass

spectrometry

AUTHOR(S): Navarro-Gonzalez, Rafael; Negron-Mendoza, Alicia;

Albarran, Guadalupe

CORPORATE SOURCE: Inst. Cienc. Nucl., UNAM, Mexico City, 04510, Mex. SOURCE: Journal of Chromatography (1991), 587(2), 247-54

CODEN: JOCRAM; ISSN: 0021-9673

DOCUMENT TYPE: Journal LANGUAGE: English

OTHER SOURCE(S): CASREACT 116:147458

AB The anal. of keto acids via reaction with 2,4-dinitrophenylhydrazine (DNP) and esterification with methanol-hydrochloric acid by gas

chromatog. and gas chromatog.-mass spectrometry is described. The derivs. formed (DNPH) are moderately stable and are easy to analyze. The separation of eighteen biol. important keto acids is described. The utility of the method in electron impact mass spectra of DNPH derivs. is presented.

AN 1992:147458 CAPLUS

DN 116:147458

OREF 116:24825a,24828a

AB The anal. of keto acids via reaction with 2,4-dinitrophenylhydrazine (DNP) and esterification with methanol-hydrochloric acid by gas chromatog. and gas chromatog.-mass spectrometry is described. The derivs. formed (DNPH) are moderately stable and are easy to analyze. The separation of eighteen biol. important keto acids is described. The utility of the method in electron impact mass spectra of DNPH derivs. is presented.

L6 ANSWER 24 OF 31 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1989:632040 CAPLUS

DOCUMENT NUMBER: 111:232040

ORIGINAL REFERENCE NO.: 111:38537a,38540a

TITLE: (R,R)-,0,0'-(2,2-Dichlorovinyl)tartaric acid: an easily synthesized optically pure vinylether. A new

synthesis of (R,R)-0,0'-diethyltartaric acid

AUTHOR(S): Uray, Georg; Lindner, Wolfgang; Reiter, Franz

CORPORATE SOURCE: Inst. Org. Chem., Karl-Franzens Univ., Graz, A-8010,

Austria

SOURCE: Synthesis (1989), (3), 194-6

CODEN: SYNTBF; ISSN: 0039-7881

DOCUMENT TYPE: Journal LANGUAGE: English

OTHER SOURCE(S): CASREACT 111:232040

GΙ

AB (R,R)-Tartaric acid diethers I (R1 = CH:CC12, Et) were prepared (R,R)-Tartaric acid reacted with CC13CHO and H2SO4 to give bis-dioxolanone II, and the latter was treated with Zn in HOAc to give I (R1 = CH:CC12). The hydrogenation at I (R1 = CH:CC12) over Pd gave I (R1 = Et).

AN 1989:632040 CAPLUS

DN 111:232040

OREF 111:38537a,38540a

IT 123848-36-2P

RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)

(preparation and ring cleavage reactions of, with methanol and ammonia)

L6 ANSWER 25 OF 31 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1989:114749 CAPLUS

DOCUMENT NUMBER: 110:114749

ORIGINAL REFERENCE NO.: 110:18921a, 18924a

TITLE: Stability and stereochemistry in the decomposition of

pentasubstituted 1-pyrazolines controlled by interactions between bulky vicinal substituents

AUTHOR(S): Nakano, Yoshihiko; Hamaguchi, Masashi; Nagai,

Toshikazu

CORPORATE SOURCE: Coll. Gen. Educ., Osaka Univ., Osaka, 560, Japan

SOURCE: Journal of Organic Chemistry (1989), 54(5), 1135-44

CODEN: JOCEAH; ISSN: 0022-3263

DOCUMENT TYPE: Journal LANGUAGE: English

OTHER SOURCE(S): CASREACT 110:114749

GΙ

AB 1,2-Diacyl-1-chloroethylenes [e.g., (Z)- and (E)-MeO2CCC1:CHCO2Me) reacted with disubstituted diazomethanes (e.g., Me2C:N2, MeCPh:N2, Ph2C:N2) to give pyrazolines and cyclopropanes (e.g., I and II, resp.). The thermal decomposition of the isolated pyrazolines was carried out. The thermal stability of the pyrazolines increased with the variation of the substituents at C5 in the order biphenylylene < Ph, Ph < Ph, Me < Me, Me, while pyrazolines bearing bulky vicinal substituents at C3, C4, and C5 in the cis configuration were substantially more stable than other isomers. This abnormal stability is explained by the reasonable expectation that bulky C4 substituents partly inhibit conformations of the conjugated substituents at C3 or C5 favorable for the decomposition, which increases the

activation energy. Although most of the thermolyses of the pyrazolines to cyclopropanes gave products with the same configuration as the starting materials, some pyrazolines bearing bulky vicinal cis groups gave mixts. of stereoisomeric cyclopropanes. The thermolysis mechanism is discussed.

AN 1989:114749 CAPLUS

DN 110:114749

OREF 110:18921a, 18924a

IT 118658-08-5P 118658-10-9P

RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)

(preparation and esterification of, with methanol)

L6 ANSWER 26 OF 31 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1988:6374 CAPLUS

DOCUMENT NUMBER: 108:6374

ORIGINAL REFERENCE NO.: 108:1215a,1218a

TITLE: Hexahydropyrroloindoles. Attempts to synthesize

2-indolyl thio ethers

AUTHOR(S): Droste, Holger; Wieland, Theodor

CORPORATE SOURCE: Max-Planck-Inst., Med. Forsch., Heidelberg, D-6900,

Fed. Rep. Ger.

SOURCE: Liebigs Annalen der Chemie (1987), (11), 901-10

CODEN: LACHDL; ISSN: 0170-2041

DOCUMENT TYPE: Journal LANGUAGE: German

OTHER SOURCE(S): CASREACT 108:6374

GΙ

AB The sensitized photochem. oxidation of tryptamines I (Boc = Me3CO2C; R = CN, CO2H, CO2Me, R1 = H; R = H, R1 = CO2H, H) gave hexahydropyrroloindoles II. The photochem. oxidation of I (R = CONH2, R1 = H) gave ketone III, whereas homotryptamine IV gave hexahydropyridoindole V. No azetidine formation from indolylglycine VI was observed Attempts to synthesize title ethers from tryptamines and sulfenyl chlorides of cysteine derivs. failed.

AN 1988:6374 CAPLUS

DN 108:6374

OREF 108:1215a,1218a

IT 10184-94-8P

RL: SPN (Synthetic preparation); PREP (Preparation) (preparation and esterification with methanol)

L6 ANSWER 27 OF 31 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1986:590448 CAPLUS

DOCUMENT NUMBER: 105:190448

ORIGINAL REFERENCE NO.: 105:30727a,30730a

TITLE: Chemistry of 2-bromo-3-(trichloromethyl)succinic

anhydride and dimethyl ester: preparation of various halogenated succinic, maleic, fumaric and malic acid

derivatives

AUTHOR(S): Nedelec, J. Y.; Blanchet, D.; Lefort, D.; Biellmann,

J. F.

CORPORATE SOURCE: Lab. Electrochim., Catal. Synth. Org., CNRS, Thiais,

94320, Fr.

SOURCE: Tetrahedron (1985), 41(20), 4503-8

CODEN: TETRAB; ISSN: 0040-4020

DOCUMENT TYPE: Journal LANGUAGE: English

OTHER SOURCE(S): CASREACT 105:190448

AB Various reactions of title adducts of BrCCl3 with maleic anhydride and

di-Me maleate or fumarate were studied. Thus, treatment of

 $\verb|erythro-MeO2CCHBrCH(CCl3)CO2Me| with AcONa in AcOH at room temperature for 24 h | \\$ 

afforded di-Me 2-(trichloromethyl)maleate, along with the corresponding

fumarate and MeO2CCHBrC(CO2Me):CC12.

AN 1986:590448 CAPLUS

DN 105:190448

OREF 105:30727a,30730a

IT 105048-49-5P 105048-50-8P

RL: SPN (Synthetic preparation); PREP (Preparation) (preparation and esterification with methanol)

L6 ANSWER 28 OF 31 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1986:442061 CAPLUS

DOCUMENT NUMBER: 105:42061

ORIGINAL REFERENCE NO.: 105:6957a,6960a

TITLE:  $\sigma$ -Skeletal rearrangement of heptalenes: thermal

transformation of heptalene-1,2-dicarboxylates into

heptalene-1,3-dicarboxylates

AUTHOR(S): Bernhard, Werner; Bruegger, Paul; Daly, John J.;

Englert, Gerhard; Schoenholzer, Peter; Hansen, Hans

CHMe<sub>2</sub>

II

Juergen

CORPORATE SOURCE: Cent. Res. Units, F. Hoffmann-La Roche and Co., Ltd.,

Basel, CH-4002, Switz.

SOURCE: Helvetica Chimica Acta (1985), 68(4), 1010-24

CODEN: HCACAV; ISSN: 0018-019X

DOCUMENT TYPE: Journal LANGUAGE: English

OTHER SOURCE(S): CASREACT 105:42061

GΙ

```
Di-Me 1,2-heptalenedicarboxylates rearrange to 1,3-dicarboxylates above
AΒ
     200°. Labeling expts. indicate that the rearrangement of I occurs
     by interchange of C-2 and C-3 in the heptalene skeleton. Thus, the I
     \rightarrow II process is the 1st thermal \sigma\text{--skeletal} rearrangement of
     heptalenes. X-ray analyses of I and II are discussed.
ΑN
     1986:442061 CAPLUS
DN
     105:42061
OREF 105:6957a,6960a
ΙT
     142-45-0
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (esterification of, with labeled methanol)
L6
     ANSWER 29 OF 31 CAPLUS COPYRIGHT 2009 ACS on STN
ACCESSION NUMBER:
                         1986:206693 CAPLUS
DOCUMENT NUMBER:
                         104:206693
ORIGINAL REFERENCE NO.: 104:32741a,32744a
                         Use of bromine in methanol - preparation of
TITLE:
                         methyl esters
                         Vairamani, M.; Rao, G. K. Viswanadha
AUTHOR(S):
CORPORATE SOURCE:
                         Mass Spectrom. Div., Reg. Res. Lab., Hyderabad, 500
                         007, India
SOURCE:
                         Indian Journal of Chemistry, Section B: Organic
                         Chemistry Including Medicinal Chemistry (1985),
                         24B(6), 691
CODEN: IJSBDB; ISSN: 0376-4699
DOCUMENT TYPE:
                         Journal
LANGUAGE:
                         English
OTHER SOURCE(S):
                         CASREACT 104:206693
     Methyl esters are prepared in 92-98\% yield and 95-98\% purity when RCO2H [R =
     PhCH2, Ph, HO2CCH2CH2, Me(CH2)n; n = 10, 12, 14] with Br2 in MeOH. The
     reaction is light and/or HBr catalyzed. An acid catalyzed mechanism for
     the reaction is indicated.
     1986:206693 CAPLUS
ΑN
     104:206693
DN
OREF 104:32741a,32744a
     Use of bromine in methanol - preparation of methyl esters
ST
     methanol esterification bromine
     Esterification
ΙT
        (of methanol-bromine as reagent for)
     7726-95-6, uses and miscellaneous
     RL: CAT (Catalyst use); USES (Uses)
        (catalyst, for esterification of carboxylic acids with methanol
ΙT
     93-58-3P
                101-41-7P
                           106-65-0P
                                         111-82-0P
                                                     112-39-0P
                                                                  124-10-7P
     RL: SPN (Synthetic preparation); PREP (Preparation)
        (preparation of, bromine-methanol reagent for)
     ANSWER 30 OF 31 CAPLUS COPYRIGHT 2009 ACS on STN
                         1985:470927 CAPLUS
ACCESSION NUMBER:
                         103:70927
DOCUMENT NUMBER:
ORIGINAL REFERENCE NO.:
                         103:11405a,11408a
TITLE:
                         Synthesis of dialkyl esters of succinic acid
AUTHOR(S):
                         Amirkhanyan, M. M.; Elanyan, M. F.
CORPORATE SOURCE:
                         USSR
                         Trudy IREA (1984), 46, 7-10
SOURCE:
                         CODEN: TKRKAM; ISSN: 0371-876X
DOCUMENT TYPE:
                         Journal
LANGUAGE:
                         Russian
OTHER SOURCE(S):
                         CASREACT 103:70927
     Direct esterification of HO2CCH2CH2CO2H (I) with 4 equiv MeOH and EtOH in
     50 mL C6H6 containing 40 g KU-2 + 8 catalyst per mol I at 110^{\circ}
     for 10 h and 140° for 6 h, resp., gave 83.2% di-Me and 96.5% di-Et
```

succinate, resp.

AN 1985:470927 CAPLUS DN 103:70927

OREF 103:11405a,11408a

IT Process optimization

(for esterification of succinic acid with methanol and

ethanol)

IT Esterification

(of succinic acid with methanol and ethanol, optimization of)

L6 ANSWER 31 OF 31 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1983:452585 CAPLUS

DOCUMENT NUMBER: 99:52585

ORIGINAL REFERENCE NO.: 99:8211a,8212a

TITLE: Reactions of cyclic anhydrides. Part IX. Facile

esterification of carboxylic acids with

organophosphorus reagents. Novel application of

alkylphosphoric esters (APE)

AUTHOR(S): Balasubramaniyan, V.; Bhatia, V. G.; Wagh, S. B. CORPORATE SOURCE: Sci. Res. Cent., H.P.T. Arts and R.Y.K. Sci. Coll.,

Nasik, 422 005, India

SOURCE: Tetrahedron (1983), 39(9), 1475-85

CODEN: TETRAB; ISSN: 0040-4020

DOCUMENT TYPE: Journal LANGUAGE: English

OTHER SOURCE(S): CASREACT 99:52585

AB The APE reagent, prepared from P4010 and excess alkanol, was used for the esterification of carboxylic acids (.apprx.50), including maleanilic,

fumaranilic, and succinanilic acids.

AN 1983:452585 CAPLUS

DN 99:52585

OREF 99:8211a,8212a

IT 16752-60-6

RL: RCT (Reactant); RACT (Reactant or reagent)
 (reagent from methanol and, for esterification)

=> select an 16 1-31 E1 THROUGH E62 ASSIGNED

=> file casreact

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FULL ESTIMATED COST 122.73 246.08

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1 "105:190448"/AN
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1 "121:280006"/AN
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1 "135:256190"/AN
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L7

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=> s methanol

L9 387445 METHANOL

=> s methanol/cn

L10 1 METHANOL/CN

=> d rn

L10 ANSWER 1 OF 1 REGISTRY COPYRIGHT 2009 ACS on STN RN 67-56-1 REGISTRY

=> file casreact

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SINCE FILE
ENTRY
SESSION
11.91
259.43

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ENTRY
SESSION
0.00
-25.42

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=> s 67-56-1 L11 163341 67-56-1

=> s 13 (1) 111 L12 181 L3 (L) L11

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=> s 112 and 17
L13 29 L12 AND L7
=> d occ 1-29
L13 ANSWER 1 OF 29 CASREACT COPYRIGHT 2009 ACS on STN
NUMBER OF HIT REACTIONS
NUMBER OF REACTIONS IN PATH
NUMBER OF REACTIONS IN SPATH
                                 COUNT
FIELD
RX(2)
ΑN
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L13 ANSWER 2 OF 29 CASREACT COPYRIGHT 2009 ACS on STN
NUMBER OF HIT REACTIONS
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NUMBER OF REACTIONS IN PATH
                                 1
NUMBER OF REACTIONS IN SPATH
                                 1
                                 COUNT
FIELD
RX(1)
ΑN
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L13 ANSWER 3 OF 29 CASREACT COPYRIGHT 2009 ACS on STN
NUMBER OF HIT REACTIONS
                                  2
NUMBER OF REACTIONS IN PATH
NUMBER OF REACTIONS IN SPATH
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FIELD
                                 COUNT
RX(1)
RX(2)
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ΑN
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L13 ANSWER 4 OF 29 CASREACT COPYRIGHT 2009 ACS on STN
NUMBER OF HIT REACTIONS
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NUMBER OF REACTIONS IN PATH
                                 1
NUMBER OF REACTIONS IN SPATH
FIELD
                                 COUNT
RX(8)
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ΑN
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L13 ANSWER 5 OF 29 CASREACT COPYRIGHT 2009 ACS on STN
NUMBER OF HIT REACTIONS
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NUMBER OF REACTIONS IN PATH
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NUMBER OF REACTIONS IN SPATH
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                                 COUNT
FIELD
RX(14)
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ΑN
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L13 ANSWER 6 OF 29 CASREACT COPYRIGHT 2009 ACS on STN
NUMBER OF HIT REACTIONS
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L13 ANSWER 7 OF 29 CASREACT COPYRIGHT 2009 ACS on STN

1

1

3

1

COUNT

NUMBER OF REACTIONS IN PATH

FIELD

RX(1)

ΑN

NUMBER OF REACTIONS IN SPATH

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NUMBER OF REACTIONS IN PATH
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NUMBER OF REACTIONS IN SPATH
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                                  COUNT
FIELD
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L13 ANSWER 8 OF 29 CASREACT COPYRIGHT 2009 ACS on STN
NUMBER OF HIT REACTIONS
NUMBER OF REACTIONS IN PATH
NUMBER OF REACTIONS IN SPATH
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RX(19)
RX(20)
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RX(23)
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RX(24)
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ΑN
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L13 ANSWER 9 OF 29 CASREACT COPYRIGHT 2009 ACS on STN
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NUMBER OF REACTIONS IN SPATH
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FIELD
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RX(1)
ΑN
L13 ANSWER 10 OF 29 CASREACT COPYRIGHT 2009 ACS on STN
NUMBER OF HIT REACTIONS
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NUMBER OF REACTIONS IN PATH
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NUMBER OF REACTIONS IN SPATH
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FIELD
                                  COUNT
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ΑN
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L13 ANSWER 11 OF 29 CASREACT COPYRIGHT 2009 ACS on STN
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NUMBER OF REACTIONS IN PATH
NUMBER OF REACTIONS IN SPATH
FIELD
                                  COUNT
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RX(3)
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ΑN
L13 ANSWER 12 OF 29 CASREACT COPYRIGHT 2009 ACS on STN
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NUMBER OF REACTIONS IN PATH
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NUMBER OF REACTIONS IN SPATH
                                  2
FIELD
                                  COUNT
RX(2)
RX(9)
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ΑN
L13 ANSWER 13 OF 29 CASREACT COPYRIGHT 2009 ACS on STN
NUMBER OF HIT REACTIONS
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NUMBER OF REACTIONS IN PATH
                                  1
NUMBER OF REACTIONS IN SPATH
                                  2
FIELD
                                  COUNT
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RX(3)
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RX(5)
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ΑN
L13 ANSWER 14 OF 29 CASREACT COPYRIGHT 2009 ACS on STN
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NUMBER OF REACTIONS IN SPATH
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FIELD
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ΑN
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L13 ANSWER 15 OF 29 CASREACT COPYRIGHT 2009 ACS on STN
NUMBER OF HIT REACTIONS
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NUMBER OF REACTIONS IN PATH
NUMBER OF REACTIONS IN SPATH
                                 COUNT
FIELD
RX(7)
RX(9)
                                 4
ΑN
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L13 ANSWER 16 OF 29 CASREACT COPYRIGHT 2009 ACS on STN
NUMBER OF HIT REACTIONS
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NUMBER OF REACTIONS IN PATH
                                 1
NUMBER OF REACTIONS IN SPATH
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FIELD
                                 COUNT
RX(18)
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ΑN
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L13 ANSWER 17 OF 29 CASREACT COPYRIGHT 2009 ACS on STN
NUMBER OF HIT REACTIONS
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NUMBER OF REACTIONS IN PATH
                                 1
NUMBER OF REACTIONS IN SPATH
                                 1
FIELD
                                 COUNT
RX(8)
                                 4
ΑN
                                 1
L13 ANSWER 18 OF 29 CASREACT COPYRIGHT 2009 ACS on STN
NUMBER OF HIT REACTIONS
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NUMBER OF REACTIONS IN PATH
                                 1
NUMBER OF REACTIONS IN SPATH
                                 1
                                 COUNT
FIELD
RX(1)
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ΑN
                                 1
L13 ANSWER 19 OF 29 CASREACT COPYRIGHT 2009 ACS on STN
NUMBER OF HIT REACTIONS
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NUMBER OF REACTIONS IN PATH
                                 1
NUMBER OF REACTIONS IN SPATH
                                 1
FIELD
                                 COUNT
RX(2)
ΑN
                                 1
L13 ANSWER 20 OF 29 CASREACT COPYRIGHT 2009 ACS on STN
NUMBER OF REACTIONS IN PATH 1
```

```
NUMBER OF REACTIONS IN SPATH
                                  2
                                  COUNT
FIELD
RX(1)
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                                  4
RX(9)
                                  4
RX(16)
ΑN
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L13 ANSWER 21 OF 29 CASREACT COPYRIGHT 2009 ACS on STN
NUMBER OF HIT REACTIONS
NUMBER OF REACTIONS IN PATH
NUMBER OF REACTIONS IN SPATH
                                  COUNT
FIELD
RX(3)
RX(6)
                                  4
RX(8)
                                  4
                                  1
ΑN
L13 ANSWER 22 OF 29 CASREACT COPYRIGHT 2009 ACS on STN
NUMBER OF HIT REACTIONS
                                  3
NUMBER OF REACTIONS IN PATH
                                  2
NUMBER OF REACTIONS IN SPATH
                                  3
FIELD
                                  COUNT
RX(23)
RX (57)
                                  3
RX(81)
                                  3
ΑN
                                  1
L13 ANSWER 23 OF 29 CASREACT COPYRIGHT 2009 ACS on STN
NUMBER OF HIT REACTIONS
                                1
NUMBER OF REACTIONS IN PATH
                                  1
NUMBER OF REACTIONS IN SPATH
                                  1
FIELD
                                  COUNT
RX(3)
                                  4
ΑN
                                  1
L13 ANSWER 24 OF 29 CASREACT COPYRIGHT 2009 ACS on STN
NUMBER OF HIT REACTIONS
                                  1
NUMBER OF REACTIONS IN PATH
                                  1
NUMBER OF REACTIONS IN SPATH
                                  1
FIELD
                                  COUNT
RX(13)
                                  4
                                  1
ΑN
L13 ANSWER 25 OF 29 CASREACT COPYRIGHT 2009 ACS on STN
NUMBER OF HIT REACTIONS
                                  3
NUMBER OF REACTIONS IN PATH
                                  2
NUMBER OF REACTIONS IN SPATH
                                  3
FIELD
                                  COUNT
RX (4)
RX(5)
                                  3
RX(14)
                                  3
ΑN
                                  1
L13 ANSWER 26 OF 29 CASREACT COPYRIGHT 2009 ACS on STN
NUMBER OF HIT REACTIONS
                          2
NUMBER OF REACTIONS IN PATH
                                  1
NUMBER OF REACTIONS IN SPATH
                                  2
```

RX(14) 3 RX(25) 3 ΑN 1 L13 ANSWER 27 OF 29 CASREACT COPYRIGHT 2009 ACS on STN NUMBER OF HIT REACTIONS 2 2 NUMBER OF REACTIONS IN PATH 2 NUMBER OF REACTIONS IN SPATH COUNT RX(3) RX(9) 4 ΑN 1 L13 ANSWER 28 OF 29 CASREACT COPYRIGHT 2009 ACS on STN NUMBER OF HIT REACTIONS 1 NUMBER OF REACTIONS IN PATH 1 NUMBER OF REACTIONS IN SPATH 1 FIELD COUNT RX(1) 3 ΑN 1 L13 ANSWER 29 OF 29 CASREACT COPYRIGHT 2009 ACS on STN NUMBER OF HIT REACTIONS 4 NUMBER OF REACTIONS IN PATH 4 NUMBER OF REACTIONS IN SPATH FIELD COUNT RX(5) RX(16) 3 RX(26) 3 3 RX(61) 1 ΑN => d ibib abs hit 1-29L13 ANSWER 1 OF 29 CASREACT COPYRIGHT 2009 ACS on STN ACCESSION NUMBER: 144:369644 CASREACT TITLE: Catalytic synthesis of dimethyl fumarate with phosphotungstic acid Li, Yangshu; Yu, Bin AUTHOR(S): CORPORATE SOURCE: Science School, Nanjing University of Technology, Nanjing, 210009, Peop. Rep. China Huagong Shikan (2004), 18(2), 57-58 SOURCE: CODEN: HUSHFT; ISSN: 1002-154X Huagong Shikan Zazhishe PUBLISHER: DOCUMENT TYPE: Journal LANGUAGE: Chinese Phosphotungstic acid was used as an esterification catalyst for AΒ synthesizing di-Me fumarate (DMF), with maleic anhydride as the starting

material and potassium bromate KBrO3 as the isomerizing agent. This method has the advantages of requiring small amount of catalyst with high catalysis activity, resulting in shorter reaction time and high DMF yield

(typically over 90%). The purification procedure of DMF is simple.

COUNT

FIELD

RX(2) OF 3 ...2 D + B ===> E

E YIELD 90%

RX(2) RCT D 67-56-1, B 110-17-8 PRO E 624-49-7 CAT 1343-93-7 Tungstate(3-), tetracosa- $\mu$ -oxododecaoxo[ $\mu$ 12-[phosphato(3-)- $\kappa$ 0: $\kappa$ 0: $\kappa$ 0: $\kappa$ 0: $\kappa$ 0': $\kappa$ 0'': $\kappa$ 0''': $\kappa$ 0'': $\kappa$ 0''': $\kappa$ 0'': $\kappa$ 0'': $\kappa$ 

L13 ANSWER 2 OF 29 CASREACT COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 142:176440 CASREACT

TITLE: Recovering of dibasic acid ester from waste alkali

liquor of caprolactam preparation

INVENTOR(S): Zhou, Xianjun; Wang, Keshun; Wu, Chonghe; Liu,

Yaozhong; Jiang, Ping

PATENT ASSIGNEE(S): Zhongxinghua Industrial Co., Ltd., Peop. Rep. China SOURCE: Faming Zhuanli Shenging Gongkai Shuomingshu, 19 pp.

CODEN: CNXXEV

DOCUMENT TYPE: Patent LANGUAGE: Chinese

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
CN 1401624	A	20030312	CN 2001-124260	20010820
CN 1172898	С	20041027		

PRIORITY APPLN. INFO.: CN 2001-124260 20010820

AB The recovering method comprises: (1) neutralizing the waste alkali liquor of caprolactam preparation to sep. aqueous phase and organic phase and oxidizing the

organic substance in the organic phase with an oxidant selected from  $\ensuremath{\mathsf{HNO3}}$ ,  $\ensuremath{\mathsf{H2O2}}$ ,

 $\rm HC104$ , or  $\rm KMnO4$ , (2) transfering the reaction product of the dibasic acid into a two-segments concentration unit to recover monobasic acid and  $\rm HNO3$  in the

first segment and decompose the residual HNO3 and nitro compound in the second segment to obtain crude C4-6 dibasic acid, (3) esterifying with C1-4 alc. in the presence of catalyst, such as H2SO4, H3PO4, HNO3, sulfonic acid, and cationic exchange resin, and distilling

RX(1) OF 1 A + B + C + 6 D ===> E + F + G

HO \* HO \* (CH<sub>2</sub>) 
$$\frac{1}{3}$$
 OH HO \* (CH<sub>2</sub>)  $\frac{1}{4}$  OH A B C

Ε

YIELD 96%

YIELD 94%

RX(1) RCT A 110-15-6, B 110-94-1, C 124-04-9, D 67-56-1 PRO E 106-65-0, F 1119-40-0, G 627-93-0 CON SUBSTAGE(1) 2 hours, 90 deg C SUBSTAGE(2) 4 hours, 120 deg C NTE yield depends on reaction conditions

AN 142:176440 CASREACT

YIELD 90%

L13 ANSWER 3 OF 29 CASREACT COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 141:259007 CASREACT

TITLE: Synthesis of chiral phosphoantigens and their activity

in  $\gamma\delta$  T cell stimulation

AUTHOR(S): Song, Yongcheng; Zhang, Yonghui; Wang, Hong; Raker,

Amy; Sanders, John; Broderick, Erin; Clark, Allen;

Morita, Craig; Oldfield, Eric

CORPORATE SOURCE: Department of Chemistry, University of Illinois at

Urbana-Champaign, Urbana, IL, 61801, USA

SOURCE: Bioorganic & Medicinal Chemistry Letters (2004),

14(17), 4471-4477

CODEN: BMCLE8; ISSN: 0960-894X

PUBLISHER: Elsevier B.V.

DOCUMENT TYPE: Journal LANGUAGE: English

AB  $\gamma\delta$  T cells expressing  $V\gamma2V\delta2$  T cell receptors are activated by a broad range of phosphorus-containing small mols., termed phosphoantigens, and are of interest in the context of the chemotherapy of B cell malignancies. Here, we report the synthesis of four pairs of chiral phosphoantigens: the bromohydrins of isopentenyl diphosphate (PhosphostimTM), the epoxides of isopentenyl diphosphate (EIPP); and the corresponding bromohydrin and epoxide analogs of but-3-enyl diphosphate. The ability of each compound to stimulate human  $V\gamma2V\delta2$  T cells was determined by TNF- $\alpha$  release and cell proliferation. In these assays, the (R)-bromohydrin diphosphates, were, on average, about twice as active as the (S)-bromohydrin diphosphates. In contrast, the (S)-form of EIPP was about twice as active as (R)-EIPP. The activities of the epoxy but -3-enyl diphosphates were both very low. These results suggest that chiral phosphoantigens, as opposed to racemic mixts., may have utility in immunotherapy.

REFERENCE COUNT:

THERE ARE 20 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

RX(1) OF 27 A + 2 B ===> C

20

C YIELD 99%

RX(1) RCT A 6236-10-8, B 67-56-1 RGT D 7719-09-7 SOC12 PRO C 81426-68-8 NTE enantiomer preparared similarly

RX(2) OF 27 A + 2 B + E ===> F...

F YIELD 84%

RX(2) RCT A 6236-10-8, B 67-56-1

STAGE (1)

RGT D 7719-09-7 SOC12

STAGE(2)

RCT E 69739-34-0 RGT G 121-44-8 Et3N

PRO F 756836-62-1

NTE enantiomer preparared similarly

AN 141:259007 CASREACT

L13 ANSWER 4 OF 29 CASREACT COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 141:191003 CASREACT

TITLE: Reactions of 2H-3,1-benzoxazine -2,4(1H)-dione

AUTHOR(S): Milea, M.; Csunderlik, C.; Fota, Daniela

CORPORATE SOURCE: Department of Organic Chemistry, The University
"POLITEHNICA" Timisoara, Timisoara, RO-1900, Rom.
SOURCE: Buletinul Stiintific al Universitatii "Politehnica"

din Timisoara Romania, Seria Chimie si Mediului

(2000), 45(1), 163-167

CODEN: BSIMFG; ISSN: 1224-6018

PUBLISHER: Universitatii "Politehnica" din Timisoara

DOCUMENT TYPE: Journal LANGUAGE: English

AB Reactions of isatoic anhydride (2H-3,1-benzoxazine -2,4(1H)-dione) with alcs. and phenols, both in the presence or absence of basic catalysts, have been studied, conditions have been established and the products characterized by physico-chemical methods. Also the reactions of isatoic anhydride with Me esters of  $\alpha$ -amino acids were studied among the reactions of isatoic anhydride with nitrogen nucleophiles.

REFERENCE COUNT: 5 THERE ARE 5 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

RX(8) OF 18 2 N + R ===> S...

2 N

$$NH_2$$
 OH

 $(8)$ 
 $S$ 
 $YIELD$  46%

RX(8) RCT N 67-56-1, R 617-45-8 RGT Q 7719-09-7 SOC12 PRO S 6384-18-5 CON 13 hours, 60 deg C AN 141:191003 CASREACT

L13 ANSWER 5 OF 29 CASREACT COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 141:54226 CASREACT

TITLE: Synthesis of 2-(5-methyl-2-phenyl-4-oxazolyl)ethanol

AUTHOR(S): Wang, Ya-Lou; Liu, Xing; Li, Jiang-Chuan CORPORATE SOURCE: Department of Medicinal Chemistry, China

Pharmaceutical University, Nanjing, 210009, Peop. Rep.

China

SOURCE: Yingyong Huaxue (2004), 21(1), 104-106

CODEN: YIHUED; ISSN: 1000-0518

PUBLISHER: Kexue Chubanshe

DOCUMENT TYPE: Journal LANGUAGE: Chinese

AB 2-(5-Methyl-2-phenyl-4-oxazolyl)ethanol (I) was synthesized from L-aspartic acid via 5-step reactions; esterification, N-benzoylation, Dakin-West reaction, cyclizaction and reduction using LiAlH4 in about 31.2% overall yield. Reacting L-aspartic acid with methanol in 0 °C gave 72.6% L-aspartic acid  $\beta$ -Me ester hydrochloride, which was benzoylated to give 82.0% N-benzoyl-L-aspartic acid  $\beta$ -Me ester (II). Dakin-West reaction of II gave 74.4% Me 3-benzamido-4-oxovalerate, which was cyclized in toluene by POCl3 to give 81.8% Me 2-(5-methyl-2-phenyl-4-oxazolyl)acetate, the latter can easily convert to I by treatment with ether solution of LiAlH4 in yield of 86%. All the compds. were characterized by 1H NMR, IR and elemental anal.

RX(14) OF 15 COMPOSED OF RX(5), RX(1), RX(2)RX(14) Q + R + B + H ===> I

I YIELD 74%

RX(5) RCT Q 56-84-8, R 67-56-1 RGT S 7719-09-7 SOC12 PRO A 16856-13-6 CON 2.5 hours, 40 deg C

RX(1) RCT A 16856-13-6, B 98-88-4

STAGE(1)

RGT D 121-44-8 Et3N SOL 75-09-2 CH2C12 CON 1.5 hours, 0 deg C

STAGE(2)

RGT E 7647-01-0 HCl SOL 7732-18-5 Water

PRO C 39741-26-9

RX(2) RCT C 39741-26-9, H 108-24-7

STAGE(1)

SOL 110-86-1 Pyridine CON 3 hours, 90 deg C

STAGE(2)

RGT E 7647-01-0 HCl SOL 7732-18-5 Water CON pH 2

PRO I 86555-45-5 AN 141:54226 CASREACT

L13 ANSWER 6 OF 29 CASREACT COPYRIGHT 2009 ACS on STN ACCESSION NUMBER: 139:6591 CASREACT

TITLE: Catalytic synthesis of dimethyl fumarate using

solid-supported superacid catalyst

AUTHOR(S): Zhao, Lifang; He, Zhusheng; Ma, Yuying

CORPORATE SOURCE: Dept. Chem .+ Chem. Eng., Baoji Coll. Arts + Sci.,

Baoji, 721007, Peop. Rep. China

SOURCE: Baoji Wenli Xueyuan Xuebao, Ziran Kexueban (2002),

22(2), 138-140 CODEN: BWZKFL

PUBLISHER: Baoji Wenli Xueyuan Xuebao Bianjibu

DOCUMENT TYPE: Journal LANGUAGE: Chinese

AB The preparation of supported catalyst, TiO2/La3+/SO42- supported on mol. sieves, and its catalytic activity to esterification of fumarate were studied. The catalyst had fine catalytic activity. The optimum conditions of the esterification were decided by orthogonal expts. as follows: activation temperature of the catalyst was 500°, the amount of catalyst was 15% (based on the mass of fumaric acid), the mole ratio of alc. to acid was 6:1 and the reaction time was 5 h. Under the optimum reaction conditions, the yield of di-Me fumarate was up to 92.3%.

RX(1) OF 1 A + 2 B ===> C

$$_{\rm HO}$$
  $_{\rm O}$   $_{\rm H_{3}C}$   $_{\rm H}$   $_{\rm A}$   $_{\rm 2~B}$   $_{\rm B}$ 

YIELD 93%

RX(1) RCT A 110-17-8, B 67-56-1

PRO C 624-49-7

CAT 7664-93-9 H2SO4, 10099-60-2 Sulfuric acid, lanthanum(3+) salt (3:2)

CON 3 hours, reflux

NTE TiO2/La/SO4 supported mol. sieves used as catalyst, optimization

AN 139:6591 CASREACT

L13 ANSWER 7 OF 29 CASREACT COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 138:337689 CASREACT

TITLE: Synthesis of dimethyl fumarate by heterogeneous

supported heteropoly acid

AUTHOR(S): Xu, Wenyuan; Peng, Daofeng; Xiong, Guoxuan; Zhu,

Xiaping

CORPORATE SOURCE: Department of Applied Chemistry, East China Institute

of Technology, Fuzhou, 344000, Peop. Rep. China

Huaxue Shiji (2002), 24(6), 367-368 SOURCE:

CODEN: HUSHDR; ISSN: 0258-3283

Huagongbu Huaxue Shiji Xinsizhan PUBLISHER:

DOCUMENT TYPE: Journal LANGUAGE: Chinese

AB Synthesis of di-Me fumarate by esterification reaction of fumaric acid with methanol catalyzed by heterogeneous supported heteropoly acid PW12/C was studied in this paper. A careful study of the effects on the

esterification reaction was done. Under these conditions, the yield of

ester was about 91.6%.

RX(1) OF 1 2 A + B ===> C

В

(1)

YIELD 91%

RX(1) RCT A 67-56-1, B 110-17-8

PRO C 624-49-7

CAT 1343-93-7 Tungstate(3-),

tetracosa- $\mu$ -oxododecaoxo[ $\mu$ 12-[phosphato(3-)-

κ0;κ0;κ0;κ0';κ0';κ

0'':κ0'':κ0'':κ0''':κ0''':κ0''']]d

odeca-, hydrogen (1:3), 7440-44-0 Carbon

CON 3 hours, reflux

138:337689 CASREACT AN

CASREACT COPYRIGHT 2009 ACS on STN L13 ANSWER 8 OF 29

ACCESSION NUMBER: 138:271016 CASREACT

TITLE: A simple, convenient and expeditious route to methyl

esters of carboxylic acids by thionyl

chloride-methanol

AUTHOR(S): Chatterjee, Tapasi; Chattopadhyay, Subhagata

CORPORATE SOURCE: Department of Chemistry, Jadavpur University, Kolkata,

700 032, India

SOURCE: Oriental Journal of Chemistry (2002), 18(2), 187-190

CODEN: OJCHEG; ISSN: 0970-020X

Oriental Scientific Publishing Co. PUBLISHER:

DOCUMENT TYPE: Journal LANGUAGE: English

A simple, convenient and expeditious preparation of 40-90% Me esters of carboxylic acids by thionyl chloride and MeOH was described. Among the 29 esters prepared were 90% 2-IC6H4CO2Me, 87% 4-MeOC6H4CO2Me and 86% Bz(CH2)2CO2Me.

REFERENCE COUNT: 15 THERE ARE 15 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

RX(19) OF 29 AM + 2 B ===> AN

ΑN YIELD 53%

RCT AM 110-15-6, B 67-56-1 RX(19) RGT D 7719-09-7 SOC12 PRO AN 106-65-0 CON 2 hours, reflux

RX(20) OF 29 ΑO + 2 B ===> AP

AP YIELD 61%

RX(20) RCT AO 87-69-4, B 67-56-1 RGT D 7719-09-7 SOC12

PRO AP 608-68-4 CON 2 hours, reflux

RX(23) OF 29 AU + 2 B ===> AV

AV YIELD 64%

RX(23) RCT AU 110-16-7, B 67-56-1 RGT D 7719-09-7 SOC12 PRO AV 624-49-7 CON 2 hours, reflux

RX(24) OF 29 AW + 2B ===> AV

AV YIELD 68%

RX(24) RCT AW 110-17-8, B 67-56-1

RGT D 7719-09-7 SOC12

PRO AV 624-49-7

CON 2 hours, reflux

AN 138:271016 CASREACT

L13 ANSWER 9 OF 29 CASREACT COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 137:352688 CASREACT

TITLE: Catalytic reaction-distillation synthesis of dimethyl

fumarate by fixed-carried heteropoly acid

AUTHOR(S): Ding, Bin; Guo, Xiangming

CORPORATE SOURCE: Jilin Institute of Chemical Technology, Jilin,

1320022, Peop. Rep. China

SOURCE: Dongbei Shida Xuebao, Ziran Kexueban (2001), 33(4),

61-65

CODEN: DSZKEE; ISSN: 1000-1832

PUBLISHER: Dongbei Shifan Daxue Xueshu Qikanshe

DOCUMENT TYPE: Journal LANGUAGE: Chinese

AB A new synthesis technol. of di-Me fumarate was presented. Fumarate, methanol, and self-made fixed-carried heteropoly acid as catalyst were used. The reaction-distillation conditions were ratio of alc. and acid about

7:1; esterification temperature about  $67-78^{\circ}$ ; and reaction time  $\leq 6$ 

(1)

h. The yield of product was up to 92%.

RX(1) OF 1 2 A + B ===> C

C YIELD 92% RX(1) RCT A 67-56-1, B 110-17-8 PRO C 624-49-7 CAT 1343-93-7D Tungstate(3-), tetracosa- $\mu$ -oxododecaoxo[ $\mu$ 12-[phosphato(3-)- $\kappa$ 0: $\kappa$ 0: $\kappa$ 0: $\kappa$ 0': $\kappa$ 

AN 137:352688 CASREACT

L13 ANSWER 10 OF 29 CASREACT COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 136:294520 CASREACT

TITLE: Synthesis of dimethyl fumarate catalyzed by SO42-/TiO2/La3+ rare earth solid superacid

AUTHOR(S): Zhou, Jianwei

CORPORATE SOURCE: Department of Chemical Engineering, Pingyuan University, Xinxiang, 453003, Peop. Rep. China

SOURCE: Henan Huagong (2001), (5), 12-14 CODEN: HEHUF3; ISSN: 1003-3467

PUBLISHER: Henansheng Shiyou Huaxue Gongye Keji Qingbao

Zhongxinzhan

DOCUMENT TYPE: Journal LANGUAGE: Chinese

AB Di-Me fumarate was synthesized from fumaric acid and methanol with S042-/Ti02/La3+ rare earth solid superacid as catalyst in dichloromethane solvent. Optimum synthetic conditions were determined: molar ratio of fumaric acid to methanol 6:1, dosage of catalyst 1.0 g/0.1 mol fumaric acid, time 4 h and solvent 25 mL. Yield of product reached above 94%.

RX(1) OF 1 2 A + B ===> C

YIELD 94%

RX(1) RCT A 67-56-1, B 110-17-8 PRO C 624-49-7 CAT 14808-79-8 Sulfate, 13463-67-7 TiO2, 215650-46-7 Sulfuric acid, lanthanum(3+) salt (3:1) SOL 75-09-2 CH2C12 AN 136:294520 CASREACT L13 ANSWER 11 OF 29 CASREACT COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 136:279092 CASREACT

TITLE: Synthesis of dimethyl fumarate from maleic acid

AUTHOR(S): Cao, Kelin

CORPORATE SOURCE: Shanxi Taiming Chemical Engineering Co., Ltd., Taigu,

030800, Peop. Rep. China

SOURCE: Huagong Jinzhan (2001), 20(4), 33-34, 39

CODEN: HUJIEK; ISSN: 1000-6613

PUBLISHER: Huaxue Gongye Chubanshe

DOCUMENT TYPE: Journal LANGUAGE: Chinese

AB Title compound was prepared from maleic acid, isomerized fumaric acid in the presence of ammonium persulfateto as catalyst, further esterification with methanol in the presence of phosphotungstic acid as catalyst, giving product with yield over 94%. The effects of catalysts and catalyst amount on the reactions were studied.

RX(2) OF 3 ...2 D + B ===> E

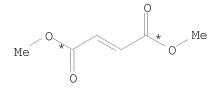
2 D

RX(2) RCT D 67-56-1, B 110-17-8 PRO E 624-49-7

CAT 12067-99-1 Tungsten hydroxide oxide phosphate

RX(3) OF 3 COMPOSED OF RX(1), RX(2)

RX(3) A + 2 D ===> E



E YIELD 94%

RX(1) RCT A 110-16-7

PRO B 110-17-8

CAT 7727-54-0 (NH4)2S208

RX(2) RCT D 67-56-1, B 110-17-8

PRO E 624-49-7

CAT 12067-99-1 Tungsten hydroxide oxide phosphate

AN 136:279092 CASREACT

L13 ANSWER 12 OF 29 CASREACT COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 136:263409 CASREACT

TITLE: Synthesis of new chiral lipophilic macrocyclic oxo

polyamines

AUTHOR(S): Yan, Qianshun; You, Jinsong; Xiang, Qingxiang; Yu,

Xiaoqi; Xie, Rugang

CORPORATE SOURCE: Department of Chemistry, Sichuan University, Chengdu,

610064, Peop. Rep. China

SOURCE: Huaxue Yanjiu Yu Yingyong (2001), 13(2), 189-191

CODEN: HYYIFM; ISSN: 1004-1656

PUBLISHER: Huaxue Yanjiu Yu Yingyong Bianjibu

DOCUMENT TYPE: Journal LANGUAGE: Chinese

AB New chiral lipophilic macrocyclic oxo polyamines were synthesized from L-glutamic acid or L-aspartic acid by esterifying with methanol, acylating with dodecanoyl chloride in chloroform in the presence of K2CO3 to obtain di-Me N-dodecanoyl-L-glutamate or L-aspartate, and transacylating with diethylenetriamine or triethylenetetraamine in methanol under refluxing. Their structures were characterized by MS, 1H NMR, and elemental anal.

RX(2) OF 15 2 A + E ===> F...

F YIELD 68%

RX(2) RCT A 67-56-1, E 56-84-8 RGT D 7719-09-7 SOC12

PRO F 6384-18-5

RX(9) OF 15 COMPOSED OF RX(2), RX(4) RX(9) 2 A + E + G ===> K

K YIELD 66%

RX(2) RCT A 67-56-1, E 56-84-8 RGT D 7719-09-7 SOC12

PRO F 6384-18-5

RX(4) RCT G 112-16-3, F 6384-18-5 RGT I 584-08-7 K2CO3 PRO K 258351-51-8

SOL 67-66-3 CHC13

AN 136:263409 CASREACT

L13 ANSWER 13 OF 29 CASREACT COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 135:256190 CASREACT

TITLE: Lipase-catalyzed synthesis of RGD diamide in aqueous

water-miscible organic solvents

AUTHOR(S): Zhang, L.-Q.; Zhang, Y.-D.; Xu, L.; Li, X.-L.; Yang,

X.-c.; Xu, G.-L.; Wu, X.-X.; Gao, H.-Y.; Du, W.-B.;

Zhang, X.-T.; Zhang, X.-Z.

CORPORATE SOURCE: Key Laboratory of Molecular Enzymology and

Engineering, Jilin University, Changchun, 130023,

Peop. Rep. China

SOURCE: Enzyme and Microbial Technology (2001), 29(2-3),

129-135

CODEN: EMTED2; ISSN: 0141-0229

PUBLISHER: Elsevier Science Ireland Ltd.

DOCUMENT TYPE: Journal LANGUAGE: English

AB 1RGD tripeptide as the cellular adhesion factor was synthesized by a combination of chemical and enzymic methods in this study. First of all, Gly-Asp diamide was synthesized by a novel chemical method in three steps including preparation of L-aspartic acid di-Me ester, chloroacetylation of L-aspartic acid di-Me ester and ammonolysis of chloroacetyl L-aspartic acid di-Me ester. Secondly, Porcine Pancreas Lipase (PPL) was used to catalyze the synthesis of Benzyl-Arg-Gly-Asp diamide in water-organic cosolvents systems. The reaction condition was optimized by examining the main factors affecting the yield of the tripeptide. The optimal reaction condition was set up as pH7.7, 15°C in 50% DMF for 8 h with the maximum yield of 76.4%. It was also found that 50% DMSO was another alternative with the tripeptide yield of 71.7%.

REFERENCE COUNT: 27 THERE ARE 27 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

RX(3) OF 6 J + 2 K ===> A...

$$^{\text{NH}_2}$$
 OH  $^{\text{H}_3\text{C}}$  OH  $^{\text{O}}$  H  $^{\text{O}}$   $^{\text{O}}$ 

Α

RX(3) RCT J 56-84-8, K 67-56-1 RGT L 7647-01-0 HC1 PRO A 6384-18-5 SOL 67-56-1 MeOH

RX(5) OF 6 COMPOSED OF RX(3), RX(1)RX(5) J + 2 K + B ===> C

С

RX(3) RCT J 56-84-8, K 67-56-1 RGT L 7647-01-0 HC1

RGT L 7647-01-0 PRO A 6384-18-5 SOL 67-56-1 MeOH

RX(1) RCT A 6384-18-5, B 79-04-9

RGT D 1310-73-2 NaOH PRO C 76385-50-7

SOL 67-66-3 CHCl3, 7732-18-5 Water

AN 135:256190 CASREACT

L13 ANSWER 14 OF 29 CASREACT COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 134:310893 CASREACT

TITLE: Synthesis of dimethyl fumarate catalyzed by composite

solid superacid SO42-/TiO2-A12O3

AUTHOR(S): Cheng, Yonghao

CORPORATE SOURCE: Department of Chemistry, Hebei Normal University,

Shijiazhuang, 050016, Peop. Rep. China

SOURCE: Riyong Huaxue Gongye (2000), 30(5), 12-13

CODEN: RHGOE8; ISSN: 1001-1803

PUBLISHER: Qinggongyebu Kexue Jishu Qingbao Yanjiuso

DOCUMENT TYPE: Journal LANGUAGE: Chinese

AB Di-Me fumarate was synthesized from fumaric acid and methanol with composite solid superacid SO42-/TiO2- Al2O3 as catalyst. Optimum

synthetic conditions were determined: methanol:fumaric acid 6:1, time 4 h, and

dosage of catalyst 3 g. Yield of product reached 91.4%.

RX(1) OF 1 2 A + B ===> C

C YIELD 91%

RX(1) RCT A 67-56-1, B 110-17-8

PRO C 624-49-7

CAT 14808-79-8 Sulfate, 13463-67-7 TiO2, 1344-28-1 Al2O3

AN 134:310893 CASREACT

L13 ANSWER 15 OF 29 CASREACT COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 133:296252 CASREACT

TITLE: A simple method for the preparation of monomethyl

esters of dicarboxylic acids by selective

esterification of the nonconjugated carboxyl group in the presence of an aromatic or conjugated carboxyl

group

AUTHOR(S): Ram, Ram N.; Meher, Nabin Kumar

CORPORATE SOURCE: Department of Chemistry, Indian Institute of

Technology, New Delhi, 110016, India

SOURCE: Journal of Chemical Research, Synopses (2000), (6),

282-283

CODEN: JRPSDC; ISSN: 0308-2342

PUBLISHER: Science Reviews Ltd.

DOCUMENT TYPE: Journal LANGUAGE: English

AB Various dicarboxylic acids were converted selectively into monomethyl esters in which the nonconjugated carboxyl group is selectively esterified in the present of an aromatic or conjugated carboxyl group at room temperature

(.apprx.25-27°) in MeOH using a catalytic amount of SOC12.

REFERENCE COUNT: 16 THERE ARE 16 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

RX(7) OF 9 2 W + 3 B ===> X + Y

HO 
$$\star$$
 OH  $\star$  O

YIELD 90%

YIELD 4%

RX(7) RCT W 97-65-4, B 67-56-1

> STAGE(1) CAT 7719-09-7 SOC12

SOL 67-56-1 MeOH

STAGE(2)

RGT E 7647-01-0 HCl SOL 7732-18-5 Water

PRO X 7338-27-4, Y 617-52-7 NTE chemoselective

RX(9) OF 9 2 AC + 3 B ===> AD + AE

В

AD YIELD 85%

AE YIELD 10%

RX(9) RCT AC 5653-88-3, B 67-56-1

STAGE(1)

CAT 7719-09-7 SOC12 SOL 67-56-1 MeOH

STAGE(2)

RGT E 7647-01-0 HCl SOL 7732-18-5 Water

PRO AD 38418-34-7, AE 80459-39-8

NTE chemoselective

AN 133:296252 CASREACT

L13 ANSWER 16 OF 29 CASREACT COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 130:66049 CASREACT

TITLE: A selective method for the preparation of aliphatic

methyl esters in the presence of aromatic carboxylic

acids

AUTHOR(S): Rodriguez, A.; Nomen, M.; Spur, B. W.

CORPORATE SOURCE: Department of Cell Biology, Univ. of Medicine and Dentistry of New Jersey, Stratford, NJ, 08084, USA

SOURCE: Tetrahedron Letters (1998), 39(47), 8563-8566

CODEN: TELEAY; ISSN: 0040-4039

PUBLISHER: Elsevier Science Ltd.

DOCUMENT TYPE: Journal LANGUAGE: English

AB 2,2-Dimethoxypropane, methanol and a catalytic amount of HCl selectively esterify aliphatic carboxylic acids in the presence of aromatic carboxylic acids

at room temperature and in high yields.

REFERENCE COUNT: 8 THERE ARE 8 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

RX(18) OF 22 AL + 3 B ===> AM

AM
YIELD 100%

RX(18) RCT AL 77-92-9, B 67-56-1 RGT D 77-76-9 Me2C(OMe)2 PRO AM 1587-20-8 CAT 75-77-4 Me3SiCl

SOL 67-56-1 MeOH

NTE chemoselective, arom. acids are not esterified

AN 130:66049 CASREACT

L13 ANSWER 17 OF 29 CASREACT COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 121:280006 CASREACT

TITLE: A convenient and mild procedure for the preparation of

hydroxy esters from lactones and hydroxy acids

AUTHOR(S): Anand, R. C.; Selvapalam, N.

CORPORATE SOURCE: Dep. Chem., Indian Inst. Technol., New Delhi, 110016,

Indi

SOURCE: Synthetic Communications (1994), 24(19), 2743-7

CODEN: SYNCAV; ISSN: 0039-7911

DOCUMENT TYPE: Journal LANGUAGE: English

AB An alc. solution of lactones and hydroxy acids stored on Amberlyst-15 results

in the formation of corresponding hydroxy esters in high yields.

$$RX(8)$$
 OF 8  $R + 2 F ===> S$ 

S YIELD 85%

ΑN

RX(8) RCT R 87-69-4, F 67-56-1

RGT D 9037-24-5 Amberlyst 15

PRO S 608-68-4 SOL 67-56-1 MeOH 121:280006 CASREACT

L13 ANSWER 18 OF 29 CASREACT COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 120:269626 CASREACT

TITLE: Catalytic synthesis of dimethyl fumarate with ferric

chloride

AUTHOR(S): Yu, Shanxin; Lei, Huanwen

CORPORATE SOURCE: Dep. Chem., Hunan Norm. UNiv., Changsha, 410081, Peop.

Rep. China

SOURCE: Huaxue Shiji (1993), 15(6), 374, 376

CODEN: HUSHDR; ISSN: 0258-3283

DOCUMENT TYPE: Journal LANGUAGE: Chinese

AB Ferric chloride (FeCl3·6H2O) can be used as a catalyst for the esterification reaction of fumaric acid instead of sulfuric acid. The conditions in synthesis of di-Me fumarate catalyzed with FeCl3·6H2O are described. The advantages of this method are: simple procedure, mild reaction conditions, non-corrosive, less pollution and purer product.

RX(1) OF 1 2 A + B ===> C

2 A

RX(1) RCT A 67-56-1, B 110-17-8 PRO C 624-49-7 CAT 7705-08-0 FeC13 AN 120:269626 CASREACT L13 ANSWER 19 OF 29 CASREACT COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 119:226419 CASREACT

TITLE: Method for preparing amino acid esters involving

(continuous) addition and distillation of alcohols

INVENTOR(S):
Takemoto, Tadashi; Takeda, Hideo

PATENT ASSIGNEE(S): Ajinomoto Co., Inc., Japan

SOURCE: Eur. Pat. Appl., 4 pp.

CODEN: EPXXDW

DOCUMENT TYPE: Patent LANGUAGE: English

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
EP 544205	A2	19930602	EP 1992-119898	19921123
EP 544205	А3	19930728		
EP 544205	В1	19950906		
R: BE, CH,	DE, FR	, GB, IT, LI,	NL	
JP 05148199	A	19930615	JP 1991-314585	19911128
CA 2084012	A1	19930529	CA 1992-2084012	19921127
US 5424476	А	19950613	US 1994-194635	19940210
PRIORITY APPLN. INFO	.:		JP 1991-314585	19911128
			US 1992-982123	19921125

AB Amino acid esters were prepared by heating a mixture of an amino acid, an alc., and H2SO4 with simultaneous addition of liquid or gaseous alc. to the reaction mixture and distillation of the same. Thus, H-Phe-OH in MeOH containing cat.

 $\rm H2SO4$  was kept at 85° for 4 h with simultaneous addition and distillation of MeOH to give 98.6% H-Phe-OMe. Simple reflux of the above reactants gave a yield of only 83-2% product.

RX(2) OF 4 2 A + E ===> F

2 A

RX(2) RCT A 67-56-1, E 56-84-8

PRO F 6384-18-5

CAT 7664-93-9 H2SO4

NTE continuous addn. and distn. of methanol

AN 119:226419 CASREACT

L13 ANSWER 20 OF 29 CASREACT COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 119:180649 CASREACT

TITLE: Preparation of pyrrolidine derivatives from amino

acids.

INVENTOR(S): Hirabayashi, Satoshi; Ike, Kazuo; Zanka, Atuhiko;

Kawakami, Takeshi; Ichihara, Masaharu

PATENT ASSIGNEE(S): Fujisawa Pharmaceutical Co., Ltd., Japan

SOURCE: PCT Int. Appl., 20 pp.

CODEN: PIXXD2

DOCUMENT TYPE: Patent LANGUAGE: Japanese

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO. KIND DATE APPLICATION NO. DATE
WO 9220652 A1 19921126 WO 1992-JP648 19920520

W: JP, US

RW: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LU, MC, NL, SE JP 04346971 A 19921202 JP 1991-219431 19910521 PRIORITY APPLN. INFO:: JP 1991-219431 19910521

OTHER SOURCE(S): MARPAT 119:180649

GΙ

was

at

AB Reaction of R2-NH-CH(CO2-R5)-CH2-CO2-R4 (R2, R4, R5 = protecting group) with R1-X [R1 = alkyl; X = leaving group) gives R2-NH-CH(CO2-R5)-CH(R1)-CO2-R4, whose reduction gives R2-NH-CH(CH2OH)-CH(R1)-CH2OH, whose condensation with H2N-R3 (R3 = protecting group) gives pyrrolidine derivs. I, which are optionally deprotected. Di-Me (S)-aspartate hydrochloride (preparation given) in CH2C12 was mixed with a saturated aqueous solution of NaHCO3, tert-di-Bu dicarbonate

added, and the resulting mixture was allowed to stand overnight to give di-Me (S)-N-tert-butoxycarbonylaspartate, which was treated with MeI in THF containing hexamethyldisilazane and BuLi at -40 to  $-30^{\circ}$  for 2 h to give, after adjusting to pH 7-8 with 1N HCl, a 1:1 mixture of threo- and erythro-(2S)-N-(tert-butoxycarbonyl)-3-methylaspartic acid di-Me ester, which was reduced with NaBH4 in THF-MeOH at  $35-50^{\circ}$  for 4 h to give a diastereomeric mixture of (2S)-N-(tert-butoxycarbonylamino)-3-methyl-1,4-butanediol, whose dimesylate ester was heated with benzylamine at  $30-50^{\circ}$  for 3 days to give 74% a diastereomeric mixture of 1-2000 for 3 days to give 74% a diastereomeric mixture of 1-2000 for 3 days to give 74% a diastereomeric mixture of 1-2000 for 3 days to give 74% a diastereomeric mixture of 1-2000 for 3 days to give 1-2000 for 3 day

35° for 2 h to give (3S,4S)-3-amino-4-methylpyrrolidine.

REFERENCE COUNT: 2 THERE ARE 2 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

$$^{\circ}$$
  $^{\circ}$   $^{\circ}$ 

● HCl

С

RX(9) OF 36 COMPOSED OF RX(1), RX(2) RX(9) A + 2 B + F ===> 
$$G$$

G

```
RGT D 7719-09-7 SOC12
PRO C 32213-95-9
SOL 67-56-1 MeOH, 68-12-2 DMF
```

RX(2) RCT F 24424-99-5, C 32213-95-9 RGT H 144-55-8 NaHCO3 PRO G 55747-84-7 SOL 75-09-2 CH2C12, 7732-18-5 Water

3 STEPS

L

Μ

RX(1) RCT A 56-84-8, B 67-56-1 RGT D 7719-09-7 SOC12 PRO C 32213-95-9 SOL 67-56-1 MeOH, 68-12-2 DMF F 24424-99-5, C 32213-95-9 RX(2) RCT H 144-55-8 NaHCO3 RGT PRO G 55747-84-7 75-09-2 CH2Cl2, 7732-18-5 Water SOL RX(3) RCT K 74-88-4, G 55747-84-7 RGT N 999-97-3 (Me3Si)2NH, O 109-72-8 BuLi PRO L 149598-19-6, M 149598-20-9

AN 119:180649 CASREACT

SOL 109-99-9 THF

L13 ANSWER 21 OF 29 CASREACT COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 117:212978 CASREACT

TITLE: Process for preparing diketopiperazine derivatives

INVENTOR(S): Toshihisa, Kato; Tadashi, Takemoto

PATENT ASSIGNEE(S): Ajinomoto Co., Inc., Japan SOURCE: Eur. Pat. Appl., 5 pp.

CODEN: EPXXDW

DOCUMENT TYPE: Patent LANGUAGE: English

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
EP 493812	A1	19920708	EP 1991-122307	19911227

R: BE, DE, FR, GB, NL

JP 04234374 A 19920824 JP 1990-418592 19901227 PRIORITY APPLN. INFO.: JP 1990-418592 19901227

OTHER SOURCE(S): MARPAT 117:212978

GΙ

Diketopiperazines I (R = C1-6-alkyl) were prepared by treating (chloroacetyl)aspartic acid with NH3, esterifying the resulting H-Gly-Asp-OH with an alkanol, and cyclizing the resulting diester neutral or weakly basic water or in an alkanol-water solvent mixture Thus, L-aspartic acid was acylated with C1CH2COCl in aqueous NaOH to give 62% C1CH2CO-L-Asp-OH, which was treated NH3 in water to H-Gly-Asp-OH. The latter was esterified with MeOH containing HCl to give 87% H-Gly-Asp(OMe)-OMe, which was cyclized in water at pH 7.0 at 60° for 3 h to give 72% I (R = Me).

RX(3) OF 10 ...F + 2 H ===> I...

Н3С─\* ОН

2 H

(3)

I YIELD 87%

RX(3) RCT F 4685-12-5, H 67-56-1 RGT J 7647-01-0 HC1 PRO I 144168-11-6 SOL 67-56-1 MeOH

RX(6) OF 10 COMPOSED OF RX(2), RX(3) RX(6) C + 2 H ===> I

I YIELD 87%

RX(2) RCT C 67036-33-3 RGT G 7664-41-7 NH3 PRO F 4685-12-5 SOL 7732-18-5 Water

RX(3) RCT F 4685-12-5, H 67-56-1 RGT J 7647-01-0 HCl PRO I 144168-11-6 RX(8) OF 10 COMPOSED OF RX(1), RX(2), RX(3)RX(8) A + B + 2 H ===> I

HO<sub>2</sub>C 
$$\xrightarrow{\text{CO}_2\text{H}}$$
  $\xrightarrow{\text{H}}$   $\xrightarrow{\text{Cl}}$   $\xrightarrow{\text{H}_3\text{C}_{\frac{1}{2}}\text{OH}}$   $\xrightarrow{\text{STEPS}}$  A B  $\xrightarrow{\text{STEPS}}$ 

I YIELD 87%

RX(1) RCT A 56-84-8, B 79-04-9 RGT D 1310-73-2 NaOH PRO C 67036-33-3 SOL 7732-18-5 Water

RX(2) RCT C 67036-33-3 RGT G 7664-41-7 NH3 PRO F 4685-12-5 SOL 7732-18-5 Water

RX(3) RCT F 4685-12-5, H 67-56-1 RGT J 7647-01-0 HC1 PRO I 144168-11-6 SOL 67-56-1 MeOH AN 117:212978 CASREACT

L13 ANSWER 22 OF 29 CASREACT COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 116:147458 CASREACT

TITLE: Analysis of keto acids as their methyl esters of 2,4-dinitrophenylhydrazone derivatives by gas

chromatography and gas chromatography-mass

spectrometry

AUTHOR(S): Navarro-Gonzalez, Rafael; Negron-Mendoza, Alicia;

Albarran, Guadalupe

CORPORATE SOURCE: Inst. Cienc. Nucl., UNAM, Mexico City, 04510, Mex. SOURCE: Journal of Chromatography (1991), 587(2), 247-54

CODEN: JOCRAM; ISSN: 0021-9673

DOCUMENT TYPE: Journal LANGUAGE: English

AB The anal. of keto acids via reaction with 2,4-dinitrophenylhydrazine (DNP)

and esterification with methanol-hydrochloric acid by gas chromatog. and gas chromatog.-mass spectrometry is described. The derivs. formed (DNPH) are moderately stable and are easy to analyze. The separation of eighteen biol. important keto acids is described. The utility of the method in electron impact mass spectra of DNPH derivs. is presented.

(23)

**(57)** 

$$RX(23)$$
 OF 90 ...2 D + AF ===> AG...

AG

$$RX(57)$$
 OF 90 2 D + AF ===> AG

AG

Α

RX(57) RCT D 67-56-1, AF 2891-14-7 RGT F 7647-01-0 HC1

PRO AG 6745-50-2 SOL 7732-18-5 Water

RX(81) OF 90 COMPOSED OF RX(22), RX(23)

RX(81) A + AE + 2 D ===> AG

ΑE

2 D

Н3С—\* ОН

RX(22) RCT A 119-26-6, AE 328-42-7 PRO AF 2891-14-7

RX(23) RCT D 67-56-1, AF 2891-14-7 RGT F 7647-01-0 HC1 PRO AG 6745-50-2 SOL 7732-18-5 Water

AN 116:147458 CASREACT

L13 ANSWER 23 OF 29 CASREACT COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 111:232040 CASREACT

TITLE: (R,R)-,0,0'-(2,2-Dichlorovinyl) tartaric acid: an easily synthesized optically pure vinylether. A new

synthesis of (R,R)-O,O'-diethyltartaric acid

AUTHOR(S): Uray, Georg; Lindner, Wolfgang; Reiter, Franz

CORPORATE SOURCE: Inst. Org. Chem., Karl-Franzens Univ., Graz, A-8010,

Austria

SOURCE: Synthesis (1989), (3), 194-6

CODEN: SYNTBF; ISSN: 0039-7881

DOCUMENT TYPE: Journal LANGUAGE: English

GΙ

AB (R,R)-Tartaric acid diethers I (R1 = CH:CC12, Et) were prepared (R,R)-Tartaric acid reacted with CC13CHO and H2SO4 to give bis-dioxolanone II, and the latter was treated with Zn in HOAc to give I (R1 = CH:CC12). The hydrogenation at I (R1 = CH:CC12) over Pd gave I (R1 = Et).

RX(3) OF 18 K + 2 L ===> M

$$C1_2C$$
 OH

 $C1_2C$  OH

 $C1_2$ 

M YIELD 68%

RX(3) RCT K 123848-33-9, L 67-56-1 RGT N 104-15-4 TsOH PRO M 123848-34-0 L13 ANSWER 24 OF 29 CASREACT COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 108:6374 CASREACT

TITLE: Hexahydropyrroloindoles. Attempts to synthesize

2-indolyl thio ethers

AUTHOR(S): Droste, Holger; Wieland, Theodor

CORPORATE SOURCE: Max-Planck-Inst., Med. Forsch., Heidelberg, D-6900,

Fed. Rep. Ger.

SOURCE: Liebigs Annalen der Chemie (1987), (11), 901-10

CODEN: LACHDL; ISSN: 0170-2041

DOCUMENT TYPE: Journal LANGUAGE: German

GΙ

The sensitized photochem. oxidation of tryptamines I (Boc = Me3CO2C; R = CN, CO2H, CO2Me, R1 = H; R = H, R1 = CO2H, H) gave hexahydropyrroloindoles II. The photochem. oxidation of I (R = CONH2, R1 = H) gave ketone III, whereas homotryptamine IV gave hexahydropyridoindole V. No azetidine formation from indolylglycine VI was observed Attempts to synthesize title ethers from tryptamines and sulfenyl chlorides of cysteine derivs. failed.

(13)

RX(13) OF 317 ...AB ===> AC...

AΒ

AC YIELD 95%

RX(13) RCT AB 10184-94-8

RGT AD 7719-09-7 SOC12, Q 67-56-1 MeOH, AE 110-86-1

Pyridine, AF 1122-58-3 4-DMAP

PRO AC 100711-44-2

SOL 67-56-1 MeOH

AN 108:6374 CASREACT

L13 ANSWER 25 OF 29 CASREACT COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 105:190448 CASREACT

TITLE: Chemistry of 2-bromo-3-(trichloromethyl)succinic

anhydride and dimethyl ester: preparation of various halogenated succinic, maleic, fumaric and malic acid

derivatives

AUTHOR(S): Nedelec, J. Y.; Blanchet, D.; Lefort, D.; Biellmann,

J. F.

CORPORATE SOURCE: Lab. Electrochim., Catal. Synth. Org., CNRS, Thiais,

94320, Fr.

SOURCE: Tetrahedron (1985), 41(20), 4503-8

CODEN: TETRAB; ISSN: 0040-4020

DOCUMENT TYPE: Journal LANGUAGE: English

AB Various reactions of title adducts of BrCCl3 with maleic anhydride and

di-Me maleate or fumarate were studied. Thus, treatment of

erythro-MeO2CCHBrCH(CCl3)CO2Me with AcONa in AcOH at room temperature for 24 h

afforded di-Me 2-(trichloromethyl)maleate, along with the corresponding

fumarate and MeO2CCHBrC(CO2Me):CC12.

RX(4) OF 19 ...K + 2 L ===> B

В

RX(4) RCT K 105048-49-5, L 67-56-1

RGT M 7664-93-9 H2SO4 PRO B 105048-44-0

RX(5) OF 19 ...N + 2 L ===> F...

H3C<del>∗</del>OH

2 L

(5)

F

Ν

RCT N 105048-50-8, L 67-56-1 RX(5)

> RGT M 7664-93-9 H2SO4 PRO F 105048-45-1

RX(14) OF 19 COMPOSED OF RX(5), RX(7)

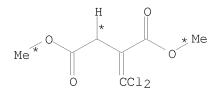
RX(14) N + 2 L ===> R

H3C<del>∗</del>OH

2 STEPS

Ν

2 L



R

RX(5) RCT N 105048-50-8, L 67-56-1

RGT M 7664-93-9 H2SO4

PRO F 105048-45-1

RX(7) RCT F 105048-45-1

RGT S 75-77-4 Me3SiCl, T 7681-82-5 NaI

PRO R 105048-47-3

SOL 75-05-8 MeCN

AN 105:190448 CASREACT

L13 ANSWER 26 OF 29 CASREACT COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 105:42061 CASREACT

TITLE:  $\sigma$ -Skeletal rearrangement of heptalenes: thermal

transformation of heptalene-1,2-dicarboxylates into

heptalene-1,3-dicarboxylates

AUTHOR(S): Bernhard, Werner; Bruegger, Paul; Daly, John J.;

Englert, Gerhard; Schoenholzer, Peter; Hansen, Hans

Juergen

CORPORATE SOURCE: Cent. Res. Units, F. Hoffmann-La Roche and Co., Ltd.,

Basel, CH-4002, Switz.

SOURCE: Helvetica Chimica Acta (1985), 68(4), 1010-24

CODEN: HCACAV; ISSN: 0018-019X

DOCUMENT TYPE: Journal LANGUAGE: English

GΙ

Di-Me 1,2-heptalenedicarboxylates rearrange to 1,3-dicarboxylates above 200°. Labeling expts. indicate that the rearrangement of I occurs by interchange of C-2 and C-3 in the heptalene skeleton. Thus, the I  $\rightarrow$  II process is the 1st thermal  $\sigma$ -skeletal rearrangement of heptalenes. X-ray analyses of I and II are discussed.

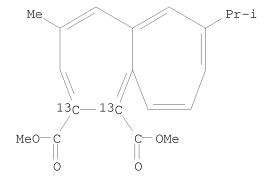
RX(14) OF 39 ...AK + AG ===> AL...

AL

RX(14) RCT AK 67-56-1, AG 101391-58-6 RGT T 7664-93-9 H2SO4 PRO AL 79519-69-0 SOL 7732-18-5 Water, 71-43-2 Benzene

$$RX(25)$$
 OF 39 COMPOSED OF  $RX(14)$ ,  $RX(15)$   
 $RX(25)$  AK + AG + H ===> AM + AN + AO

AN



ΑO

RX(14) RCT AK 67-56-1, AG 101391-58-6

RGT T 7664-93-9 H2SO4

PRO AL 79519-69-0

SOL 7732-18-5 Water, 71-43-2 Benzene

RX(15) RCT H 489-84-9, AL 79519-69-0

PRO AM 101391-59-7, AN 101391-60-0, AO 101391-61-1

SOL 119-64-2 Tetralin

AN 105:42061 CASREACT

L13 ANSWER 27 OF 29 CASREACT COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 104:206693 CASREACT

TITLE: Use of bromine in methanol - preparation of methyl

esters

AUTHOR(S): Vairamani, M.; Rao, G. K. Viswanadha

CORPORATE SOURCE: Mass Spectrom. Div., Reg. Res. Lab., Hyderabad, 500

007, India

SOURCE: Indian Journal of Chemistry, Section B: Organic

Chemistry Including Medicinal Chemistry (1985),

24B(6), 691

CODEN: IJSBDB; ISSN: 0376-4699

DOCUMENT TYPE: Journal LANGUAGE: English

AB Methyl esters are prepared in 92-98% yield and 95-98% purity when RCO2H [R = PhCH2, Ph, HO2CCH2CH2, Me(CH2)n; n = 10, 12, 14] with Br2 in MeOH. The reaction is light and/or HBr catalyzed. An acid catalyzed mechanism for the reaction is indicated.

(3)

RX(3) OF 9 2 A + G ===> H

G

Η

RX(3) RCT A 67-56-1, G 110-15-6

RGT D 7726-95-6 Br2 PRO H 106-65-0

SOL 67-56-1 MeOH

NTE best yield in light

RX(9) OF 9 G + 2 A ===> H

Н3С С № Н

2 A

(9)

Η

G

RX(9) RCT G 110-15-6, A 67-56-1

RGT O 10035-10-6 HBr

PRO H 106-65-0

SOL 67-56-1 MeOH, 7732-18-5 Water

AN 104:206693 CASREACT

L13 ANSWER 28 OF 29 CASREACT COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 103:70927 CASREACT

TITLE: Synthesis of dialkyl esters of succinic acid

AUTHOR(S): Amirkhanyan, M. M.; Elanyan, M. F.

CORPORATE SOURCE: USSR

SOURCE: Trudy IREA (1984), 46, 7-10 CODEN: TKRKAM; ISSN: 0371-876X

DOCUMENT TYPE: Journal LANGUAGE: Russian

AB Direct esterification of HO2CCH2CH2CO2H (I) with 4 equiv MeOH and EtOH in 50 mL C6H6 containing 40 g KU-2 + 8 catalyst per mol I at 110° for 10 h and 140° for 6 h, resp., gave 83.2% di-Me and 96.5% di-Et succinate, resp.

RX(1) OF 4 A + 2 B ===> C

C YIELD 83%

RX(1) RCT A 110-15-6, B 67-56-1 PRO C 106-65-0 AN 103:70927 CASREACT

L13 ANSWER 29 OF 29 CASREACT COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 99:52585 CASREACT

TITLE: Reactions of cyclic anhydrides. Part IX. Facile

esterification of carboxylic acids with

organophosphorus reagents. Novel application of

alkylphosphoric esters (APE)

AUTHOR(S): Balasubramaniyan, V.; Bhatia, V. G.; Wagh, S. B. CORPORATE SOURCE: Sci. Res. Cent., H.P.T. Arts and R.Y.K. Sci. Coll.,

Nasik, 422 005, India

SOURCE: Tetrahedron (1983), 39(9), 1475-85

CODEN: TETRAB; ISSN: 0040-4020

DOCUMENT TYPE: Journal LANGUAGE: English

AB The APE reagent, prepared from P4010 and excess alkanol, was used for the esterification of carboxylic acids (.apprx.50), including maleanilic, fumaranilic, and succinanilic acids.

RX(5) OF 73 L + 2 H ===> M

M YIELD 72%

RX(5) RCT L 6915-18-0, H 67-56-1 RGT D 16752-60-6 P4010

PRO M 23055-10-9

RX(16) OF 73 AC + 2 H ===> AD

AD YIELD 60%

RX(16) RCT AC 110-15-6, H 67-56-1 RGT D 16752-60-6 P4010 PRO AD 106-65-0

RX(26) OF 73 AN + 2 H ===> AS

AS YIELD 68%

RX(26) RCT AN 7407-59-2, H 67-56-1 RGT D 16752-60-6 P4010

PRO AS 84569-14-2

RX(61) OF 73 L + 2 H  $\Longrightarrow$  M

M YIELD 73%

RX(61) RCT L 6915-18-0, H 67-56-1

RGT D 16752-60-6 P4010

PRO M 23055-10-9

AN 99:52585 CASREACT